

STOICHIOMETRY PROBLEMS

1. The combustion of a sample of butane, C_4H_{10} (lighter fluid), produced 2.46 grams of water.



- (a) How many moles of water formed?
- (b) How many moles of butane burned?
- (c) How many grams of butane burned?
- (d) How much oxygen was used up in moles?
- (e) How much oxygen was used up in grams?

2. Sodium metal burns in air according to the balanced reaction below.



- (a) How many moles of oxygen are needed to react with 9.5 g of sodium completely?
- (b) How many grams of sodium are needed to produce 12.5 g of sodium oxide?

3. In the combustion of 54.5 g of butane (C_4H_{10}), how many grams of CO_2 are produced? Write and balance the equation before solving.

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web jun 25 2019 fixed obligations as with other legal systems under islamic law the deceased s estate must first be used to pay funeral expenses debts and other obligations what remains is then divided amongst heirs the quran says of what they leave after any bequest they may have made or debt 4 12 writing a will

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web may 21 2020 the islamic law of inheritance is based on the principle of fair and equitable division among heirs however there are some qur anic verses that specify different shares for different relatives for example a husband is entitled to a double share of his wife s inheritance and a parent is entitled to a double share of their child s