

Electrolytes Worksheet

1. What is an electrolyte? Give an example.

substance that, when dissolved in water, conducts electricity

2. Classify the following substances by type of electrolyte (acid, base or salt).

KOH	<u>B</u>	Ba(NO ₃) ₂	<u>S</u>	KF	<u>S</u>
H ₂ SO ₃	<u>A</u>	HNO ₃	<u>A</u>	Na ₂ CO ₃	<u>S</u>
Mg(OH) ₂	<u>B</u>	NH ₄ OH	<u>B</u>	Fe(OH) ₃	<u>B</u>
HCl	<u>A</u>	MgCl ₂	<u>S</u>	Ca(OH) ₂	<u>B</u>

3. Write the equation for the electrolytic dissociation of the following compounds:



4. To check the electrical conductivity of certain substances, a student used a conductivity apparatus equipped with a light bulb. Her observations are listed in the following table. Which one of the following groups of substances contains only electrolytes?

Substances	Observations
HCl ✓	Bright light
CH ₃ OH	No light
MgCl ₂ ✓	Faint light
NaOH ✓	Bright light
CH ₃ COOH ✓	Faint light
CCl ₄	No light

- A) CH₃OH and CCl₄ C) CH₃OH, NaOH and CH₃COOH
 B) HCl, MgCl₂ and CCl₄ D) HCl, MgCl₂, NaOH and CH₃COOH

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