

# Quality Control Interview Questions & Answers

**Redox titrations:** Where the titrant is oxidizing agents or reducing agents.

**Q47. What is the Ultraviolet (UV) and visible spectroscopy range?**

The range of UV Spectroscopy is 200-400 nm, and visible spectroscopy ranges from 400- 800 nm.

**Q48. What is the use of UV Spectroscopy?**

Spectroscopy can be used to detect functional groups, and impurities, and perform qualitative and quantitative analyses.

**Q49. What is meant by the solution?**

A solution is a mixture of liquids, gases, and solids, the solution consists of many different types of solutes like salts, oxygen, and organic molecules.

**Q50. Describe the saturated and unsaturated solutions**

A saturated solution is defined as a solution in which a solvent is not capable of dissolving any more solute at a given temperature.

At a given temperature, an unsaturated solution is one in which the solvent is Capable of dissolving any extra solute.

**Q51. What is the difference between qualitative and quantitative analysis?**

Qualitative analysis involves the identification of the compound or chemical based on their chemical (absorption, emission) or physical properties, e.g., melting point and boiling point.

Quantitative analysis: this involves estimation or determination of the concentration or amount of the chemical compounds or components.

**Q52. Explain the principle of ultraviolet spectroscopy?**

Ultraviolet spectroscopy uses light in the UV part of the electromagnetic spectrum. UV absorption spectra form when the outer electrons of a molecule or an atom absorb energy and move from a lower to a higher energy level. The wavelength absorbance of each molecule is unique.

**Q53. Define Molarity?**

A number of moles of solute per litre solution. Molarity is denoted with a capital "M".

**Q54. Define molality?**

The number of moles of solute per kilogram solvent. It is denoted with a small "m".

**Q55. Define normality?**

The number of moles equivalent per litre solution.

**Q56. What are buffer solutions?**

A buffer solution is an aqueous solution consisting of a mixture of a weak acid and its conjugate base or vice versa.

# Document Control Interview Questions And Answers

**Donald K. Burleson**



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