

Question number	Answer	Marks	Guidance
1 (a)	$n(\text{NaOH}) = 0.125 \times 22.40/1000 = 2.80 \times 10^{-3} \text{ mol}$	B1	
1 (b)	$n(\text{NaOH}) = 0.5 \times 2.80 \times 10^{-3} = 1.40 \times 10^{-3} \text{ mol}$	B1	
1 (c)	$c = 1.40 \times 10^{-3} \times 1000/25.0 = 0.0560 \text{ mol dm}^{-3}$	B1	
2	N in NH_3 has been oxidised from -3 to 0 in N_2 Cu in CuO has been reduced from $+2$ to 0 in Cu	B1 B1	
3 (a) (i)	Mg has lost 2 electrons and has been oxidized Fe has lost 3 electrons and has been reduced	B1 B1	
3 (a) (ii)	iron(III) nitrate(V)	B1	
3 (b) (i)	Mn has been reduced from $+4$ in MnO_2 to $+2$ in MnCl_2	B2	
3 (b) (ii)	In HCl , Cl has oxidation number of -1 . Cl has been oxidised from -1 in HCl to 0 in Cl_2 Cl in HCl has an oxidation number of -1 which is unchanged in MnCl_2	B1 B1	
4 (a) (i)	For each equation, 1 mark for balanced equation, 1 mark for state symbols. $\text{CaCO}_3(\text{s}) + 2\text{HCl}(\text{aq}) \rightarrow \text{CaCl}_2(\text{aq}) + \text{CO}_2(\text{g}) + \text{H}_2\text{O}(\text{l})$ $\text{CaO}(\text{s}) + 2\text{HCl}(\text{aq}) \rightarrow \text{CaCl}_2(\text{aq}) + \text{H}_2\text{O}(\text{l})$ $\text{Ca}(\text{OH})_2(\text{aq}) + 2\text{HCl}(\text{aq}) \rightarrow \text{CaCl}_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$	B1 x 2 B1 x 2 B1 x 2	
4 (a) (ii)	An acid has been neutralised by a base to form water	B1	
4 (a) (iii)	$\text{H}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l})$	B1	
4 (b) (i)	Reduction is increase in oxidation number and gain of electrons Oxidation is decrease in oxidation number and loss of electrons	B1 B1	
4 (b) (ii)	$\text{Ca}(\text{s}) + 2\text{HCl}(\text{aq}) \rightarrow \text{CaCl}_2(\text{g}) + \text{H}_2(\text{g})$	B2	
4 (b) (iii)	Ca has been oxidised from 0 to $+2$ H has been reduced from $1+$ to 0 .	B1 B1	
5 (a)	$\text{WO}_3 + 3\text{H}_2 \rightarrow \text{W} + 3\text{H}_2\text{O}$	B1	

Access Chapter 4 Answers

**Emmett Dulaney, Chuck Easttom, James
Michael Stewart, S. Russell Christy**

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networks have significantly facilitate our human life securing and ensuring high quality services of these networks are still a primary concern In particular anomalous protocol operation in wireless multi hop networks has recently received considerable attention in the research community These relevant security issues are fundamentally different from those of wireline networks due to the special characteristics of wireless multi hop networks such as the limited energy resources and the lack of centralized control These issues are extremely hard to cope with due to the absence of trust relationships between the nodes To enhance security in wireless multi hop networks this dissertation addresses both MAC and routing layers misbehaviors issues with main focuses on thwarting black hole attack in proactive routing protocols like OLSR and greedy behavior in IEEE 802.11 MAC protocol Our contributions are briefly summarized as follows As for black hole attack we analyze two types of attack scenarios one is launched at routing layer and the other is cross layer We then provide comprehensive analysis on the consequences of this attack and propose effective countermeasures As for MAC layer misbehavior we particularly study the adaptive greedy behavior in the context of Wireless Mesh Networks WMNs and propose FLSAC Fuzzy Logic based scheme to Struggle against Adaptive Cheaters to cope with it A new characterization of the greedy behavior in Mobile Ad Hoc Networks MANETs is also introduced Finally we design a new backoff scheme to quickly detect the greedy nodes that

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web the checklist over the following 2 pages allows a subjective overview of a child s pragmatic skills it is often useful to have a parent carer fill out the checklist as well as a preschool childcare school teacher as a child s pragmatic skills may vary in mastery between contexts

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web mcginnis pragmatic skills checklist what are pragmatic skills pragmatic skills facilitate our social interactions we use pragmatics to get various social communication accomplished we attend request tell clarify we also adjust our messages based on our knowledge of the situation and the participants involved

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