

# Chemistry Study Guide

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## Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯 ➡.

## Matter:

Anything that has **mass** and **occupies space** ✨🌍

Examples: air, gases, liquids, solidz!

## Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉.

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	$p^+$	1	+1	Nucleus
Neutron	$n$	1	0	Nucleus
Electron	$e^-$	$\sim 0$	-1	Orbit/Shell

# General Chemistry Study Guide

**Yijin Wang**



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Hurd,1928 **Student Study Guide for Principles of General Chemistry** Martin Silberberg,2012-04-02 Designed to help students recognize their learning style understand how to read classify and create a problem solving list and practice problem solving skills each chapter provides study objectives and a summary of the text followed by sample problems with detailed solutions as well as true false questions and a self test with all answers provided at the end of the chapter

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level courses The Author of this book Dr Mark Bill has an extensive teaching experience in chemistry which makes this book much helpful

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*General Chemistry I, Student Study Guide* Enrique Olivas, 2015-08-26

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*All Hands*, 1968

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