

THE MOLE: LAB ACTIVITY

Station 1:

A) Determine the mass of one drop of water by measuring the mass of 50 drops of water.

Mass of dish (or beaker)	= _____ g
Mass of dish (or beaker) + 50 drops water	= _____ g
Mass of 50 drops water	= _____ g
Mass of 1 drop of water	= _____ g

B) How many molecules are there in one drop of water?

Remember 1 mole of a substance is 6.02×10^{23} particles (atoms or molecules).
Therefore, 6.02×10^{23} molecules of water will weigh 18.02 g/mol

1. Find # of moles of water for 1 drop $n = (\text{mass})/(\text{Molar Mass})$
2. # of molecules in one drop of water $= n_{\text{water}} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole}}$
 $= ? \text{ molecules}$

Station 2

Calculate the number of copper atoms in a penny.

1. Find # of moles of Cu $n = (\text{mass})/(\text{Molar Mass})$
2. # of Cu atoms in a penny $= n_{\text{Cu}} \times \frac{6.02 \times 10^{23} \text{ atoms}}{1 \text{ mole}}$
 $= ? \text{ atoms}$

Station 3

A. Calculate the number of moles in one spoon of sugar. Assume that sugar is made up entirely of sucrose molecules ($\text{C}_{12}\text{H}_{22}\text{O}_{11}$).

$$M_{\text{sucrose}} = 342.34 \text{ g/mol}$$

$$n_{\text{sugar}} = \frac{M_{\text{sucrose}}}{M_{\text{sucrose}}}$$
$$= \underline{\hspace{2cm}} \text{ mol}$$

B. Calculate the number of molecules of sugar in one spoon of sugar.

$$\begin{aligned} \# \text{ of molecules sucrose} &= n_{\text{sucrose}} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mol}} \\ &= \underline{\hspace{2cm}} \text{ mol} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole}} \\ &= ? \text{ molecules} \end{aligned}$$

Mole Lab Counting And Weighing Answers

Didier Musso



Mole Lab Counting And Weighing Answers:

Chemistry Essentials For Dummies John T. Moore, 2019-04-16 Chemistry Essentials For Dummies 9781119591146 was previously published as Chemistry Essentials For Dummies 9780470618363 While this version features a new Dummies cover and design the content is the same as the prior release and should not be considered a new or updated product Whether studying chemistry as part of a degree requirement or as part of a core curriculum students will find Chemistry Essentials For Dummies to be an invaluable quick reference guide to the fundamentals of this often challenging course Chemistry Essentials For Dummies contains content focused on key topics only with discrete explanations of critical concepts taught in a typical two semester high school chemistry class or a college level Chemistry I course from bonds and reactions to acids bases and the mole This guide is also a perfect reference for parents who need to review critical chemistry concepts as they help high school students with homework assignments as well as for adult learners headed back into the classroom who just need to a refresher of the core concepts The Essentials For Dummies Series Dummies is proud to present our new series The Essentials For Dummies Now students who are prepping for exams preparing to study new material or who just need a refresher can have a concise easy to understand review guide that covers an entire course by concentrating solely on the most important concepts From algebra and chemistry to grammar and Spanish our expert authors focus on the skills students most need to succeed in a subject

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Unveiling the Magic of Words: A Report on "**Mole Lab Counting And Weighing Answers**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Mole Lab Counting And Weighing Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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