

# Chemistry

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(Chapter 11) (Alcohols Phenols and Ethers) (Exercises)

(Class 12)



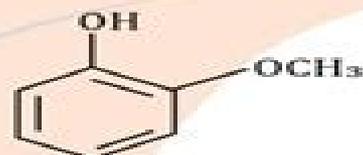
### Question 11.15:

Explain why is *ortho* nitrophenol more acidic than *ortho* methoxyphenol?

Answer 11.15:



*o* - Nitrophenol



*o* - Methoxyphenol

The nitro-group is an electron-withdrawing group. The presence of this group in the *ortho* position decreases the electron density in the O-H bond. As a result, it is easier to lose a proton. Also, the *o*-nitrophenoxide ion formed after the loss of protons is stabilized by resonance. Hence, *ortho* nitrophenol is a stronger acid.

On the other hand, methoxy group is an electron-releasing group. Thus, it increases the electron density in the O-H bond and hence, the proton cannot be given out easily.

For this reason, *ortho*-nitrophenol is more acidic than *ortho*-methoxyphenol.

### Question 11.16:

Explain how does the -OH group attached to a carbon of benzene ring activate it towards electrophilic substitution?

Answer 11.16:

The -OH group is an electron-donating group. Thus, it increases the electron density in the benzene ring as shown in the given resonance structure of phenol.



As a result, the benzene ring is activated towards electrophilic substitution.

### Question 11.17:

Give equations of the following reactions:

- (i) Oxidation of propan -1-ol with alkaline  $\text{KMnO}_4$  solution.
- (ii) Bromine in  $\text{CS}_2$  with phenol.
- (iii) Dilute  $\text{HNO}_3$  with phenol.
- (iv) Treating phenol with chloroform in presence of aqueous  $\text{NaOH}$ .

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# Chapter 12 Chemistry

**Joacim Rocklöv**



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**Hearings and Reports on Atomic Energy** United States. Congress. Joint Committee on Atomic Energy,1956

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nearby He has his suspicions on the head administrator on the disappearance of his father and the demise of their school  
However the girls are already wrapped up with a recent turn of event on a string of botched alcoholic beverages more  
commonly linked in Ireland Can the girls solve this fatal case and help their friend      Baccharis Geraldo Wilson  
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**An Introduction to Organic Chemistry** Alexander Lowy, Benjamin Harrow, 1924

**Student Solutions Manual to Accompany Chemistry** William Joseph Pietro, 1990

**The High School Journal** ,1921 *Integrated Physics and Chemistry, Chapter 12, Activities* Paradigm Accelerated Curriculum, 2005-01-01 Key topics speed energy force simple machines Laws of Motion heat pressure density wave motion light electricity circuits current power safety with electricity discovery by design careers in physics Newton Franklin IPC consists of twelve chapters of text and twelve companion student activity books This course introduces students to the people places and principles of physics and chemistry It is written by internationally respected scientist author John Hudson Tiner who applies the vignette approach which effectively draws readers into the text and holds attention The author and editors have deliberately avoided complex mathematical equations in order to entice students into high school level science Focus is on the people who contributed to development of the Periodic Table of the Elements Students learn to read and apply the Table while gaining insight into basic chemistry and physics This is one of our most popular courses among high school students especially those who have a history of under performance in science courses due to poor mathematical and reading comprehension skills The course is designed for two high school transcript credits Teachers may require students to complete all twelve chapters for two transcript credits or may select only six chapters to be completed for one transcript credit for Physical Science Physics or Chemistry Compliance with state and local academic essential elements should be considered when specific chapters are selected by teachers As applicable to local policies transcript credit may be assigned as follows when students complete all 12 chapters Physical Science for one credit and Chemistry for one credit or Integrated Physics and Chemistry for two credits May require supplemental local classes labs

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**The Independent** ,1914

**Science for the School and Family: Mineralogy and geology** Worthington Hooker, 1894

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Chemical Abstracts ,1925

Catalogue Johnson, George P., bookseller, Edinburgh, 1881

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