

$$r_{11} = \left[ \frac{k}{k-1} \right] \left( 1 - \frac{\sum \sigma_b^2}{\sigma_t^2} \right)$$

Dimana,  $r_{11}$  = reliabilitas instrumen (koefisien *Alpha Cronbach*)

$k$  = jumlah butir pertanyaan dalam instrumen

$\sum \sigma_b^2$  = jumlah varians butir-butir pertanyaan

$\sigma_t^2$  = varians total

# Rumus Alpha Cronbach Arikunto

**Hongru Du**



**Rumus Alpha Cronbach Arikunto:**

## The Enigmatic Realm of **Rumus Alpha Cronbach Arikunto**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Rumus Alpha Cronbach Arikunto** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of people who partake in its reading experience.

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objective to correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objects can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound s normal boiling point refers to its boiling point at a pressure of 760 text mm ce hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds

to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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dec 7 2019 1987 the sample essay on boiling point determination lab report deals with a framework of research based facts approaches and arguments concerning this theme to see the essay s introduction body paragraphs and conclusion read on should be based on observations made during the experiment

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

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there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of material

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jan 7 2020 grâce à la finesse de sa grille le chinois permet de filtrer un bouillon une sauce ou un coulis pour les débarrasser de toutes traces de morceaux de légumes de viande ou de

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on utilise généralement le chinois pour passer une préparation liquide cette action consiste à extraire le liquide d'une préparation en l'appuyant fortement on préférera utiliser un chinois entièrement métallique et inoxydable pour s'adapter à toutes les préparations le chinois est très pratique pour filtrer un

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la cuisine chinoise est une des plus réputées au monde et sans doute celle qui comporte le plus de variations elle se répartit traditionnellement en quatre familles cuisine du dongbei cuisine shanghaienne cuisine sichuanaise et cuisine cantonaise la cuisine chinoise est liée à la société à la philosophie et à la médecine

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les spécialités sont diverses et varient selon les régions du pays en chine les préparations sont toujours réalisées minutieusement dans les règles de l art la cuisine chinoise complète et variée une cuisine harmonieuse et très savoureuse

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on se dit il qu il y a surement un tas de trucs et astuces à connaître pour pouvoir cuisiner chinois c est pas faux table des matières trucs et astuces ingrédients indispensables trucs et astuces 1 la sauce soja ou 酱油 jiang you trucs et astuces 2 l huile de sésame 芝麻 zhi ma you trucs et astuces 3 5 épices 五香 wu xiang fen

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