

## Technical Manual

# JBL EON518S/230

## SPECIFICATIONS



### ACOUSTIC & ELECTRICAL SPECIFICATIONS:

- System Power Rating: 500W continuous, 1000W peak
- Frequency Response: 42 Hz – 100 Hz (+3 dB)
- Frequency Range: 38 Hz – 130 Hz (-10 dB)
- Input Gain: - infinity to +4 dBu
- Input Impedance: Nominal 20K ohms (balanced)
- Maximum SPL: 129 dB, 1 watt @ 1 meter
- Crossover Frequency: 120 Hz (HPF switchable on outputs)

### SYSTEM COMPONENTS:

- Cabinet: Enclosure (Not for Sale)
- Baffle/Grille Assembly: 441958-001
- Low Frequency Transducer: 208G, 400 mm (16 in) Differential Drive® integrated woofer with dual voice coils, 4 ohm impedance
- DC Resistance: 2.00 ohm ± 10%
- Input Connectors: Balanced XLRs with loop through

### SYSTEM COMPONENTS: (continued)

- XLR Pin Assignments: (stereo) to balance XLR satellite outputs, 1/4" Speaker level input
- Suspension/ Mounting: Pin 2 (+), Pin 3 (-), Pin 1 (GND) 38 mm pole socket on the top

### AURAL SWEEP TEST SPECIFICATIONS:

- A. System Aural Sweep Test: 500mV Input, 20 Hz to 500 Hz Sweep Duration: 5 seconds
- B. L.F. Aural Sweep Test: 3.0V Input, 10 Hz to 300 Hz

### PHYSICAL SPECIFICATIONS:

- Enclosure Dimensions: 617 x 569 x 652 mm D (24.3 x 22.4 x 25.7 in D)
- Shipping Weight: 65 lb (29.5 kg.)

### WARRANTY INFORMATION:

- Refer to Warranty Statement packed with each product.

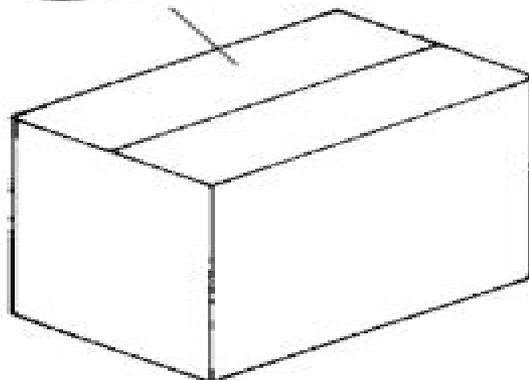
## PACKAGE

Top / Bottom In net, (2)  
304817-001

PKG, Microfoam  
49790P  
20" x 18" sheets

Carton  
304818-001

Cord, AC, Line  
425-00000-00 120V  
425-00000-01 230V

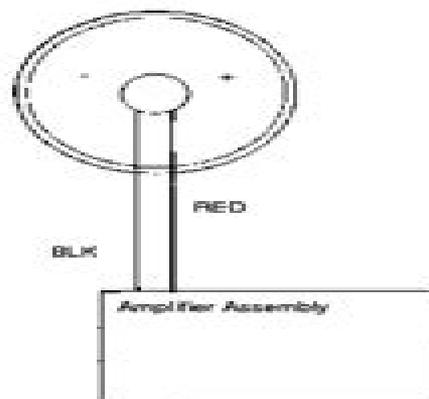


Warranty 50660

Owner's Manual  
304822-001  
(Available on  
web-site only)

## WIRING DIAGRAM

Driver, Low Freq  
208G  
(30-4481-001X)  
Cone Papl  
C-4R268



# Jbl Eon 518s Service Manual

**G Orfield**



**Jbl Eon 518s Service Manual:**

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web suggested activity use this resource could be used as a whole class investigation with older children observing differences and similarities with hand warmers leading to research on how they work children can investigate why materials have been chosen for the hand warmers linking to previous knowledge on materials and properties

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web  $m \times c \times \Delta t$  equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and  $\Delta t$  is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of *guided inquiry designing the ideal hand warmer - Dec 27 2022*

web the heat of reaction  $\Delta H_{\text{soln}}$  is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as  $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$   $\Delta H_{\text{soln}} = -44.2 \text{ kJ mol}^{-1}$  rxn write equations to similarly represent the dissolving process for

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web hand warmers are usually placed inside the gloves to keep your hands warm the heat keeping your hands warm is formed as a result of an exothermic reaction which occurs in a small packet hand warmers have 5 main ingredients in them salt carbon polypropylene vermiculite water and iron

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web safety 1 all chemicals in the laboratory are to be considered dangerous do not touch taste or smell any chemicals unless specifically instructed to do so 2 never remove chemicals or other materials from the laboratory area 3 do not eat food drink beverages or chew gum in the laboratory

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web when the hand warmer pack is activated the solid dissolves in water and produces a large temperature change in the form of heat in this laboratory experiment the heat of solution will be calculated from the data for six salts in order to

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web background the hand warmers are filled with sodium acetate click the metal disc and you will notice the liquid changes state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

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web oct 9 2021 sodium acetate coin or other small piece of metal plastic bags seal the coin inside a small bag place this bag inside a larger bag containing supersaturated sodium acetate get this by dissolving as much sodium acetate as possible in hot water this chemical hand warmer is reusable

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution  $\Delta H_{\text{soln}}$  or the amount of energy released or absorbed in kJ mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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web feb 2 2017 this is a video outlining a flinn lab on designing a hand warmer it was created by alex brinley charis conwell and siena joy for our ap chemistry class

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web hand warmers provide a unique and fun way to study the chemistry of crystal formation and heat generation by pressing a button in a pouch which contains a supercooled solution you start a rapid exothermic heat producing crystallization in this

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web lab investigation in today s experiment you will design and execute an experimental procedure to determine which of three ionic compounds is most suitable for use in a hand warmer the three compounds you will use is determined by the group number you are assigned the ideal hand warmer

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