

Name: \_\_\_\_\_Armando Duran\_\_\_\_\_ Date: \_\_\_\_\_9/24/2021\_\_\_\_\_

## Student Exploration: Osmosis

**Vocabulary:** cell membrane, concentration, diffusion, dynamic equilibrium, osmosis, semipermeable membrane, solute, solvent

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. Suppose you were trapped on a desert island with no sources of fresh water. Should you drink water from the ocean? Explain why or why not.

**No because the salt water is dangerous, and the water is dirty.**

2. What do you think would happen if you watered your houseplants with salt water?

**I think they would die slowly, with no chance of recovery.**

### Gizmo Warm-up

A **cell membrane** is a thin "skin" that surrounds a cell. It is a **semipermeable membrane**, which means that some particles pass through the membrane easily while others cannot.

The Osmosis Gizmo portrays a cell (red square) in a solution of purple **solute** particles dissolved in green **solvent** particles. Press **Play** (▶) and observe.

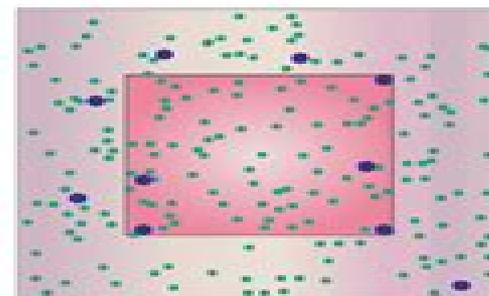
1. Which particles can pass through the cell membrane?

\_\_\_\_\_ **Solvent can pass through the cell membrane**

2. Which particles cannot pass through the cell membrane? \_\_\_\_\_ **Solute can't pass through the cell membrane** \_\_\_\_\_

3. Click **Reset** (↺), and then click **Play** again. What do you notice about the size of the cell?

\_\_\_\_\_ **the cell gets bigger than smaller like a pulse.**



# Explore Learning Student Exploration Osmosis Answer Key

**Yi-Tong Ma**



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