

PERIODIC TABLE OF THE ELEMENTS

Table of Selected Radioactive Isotopes

GROUP 1IA		3BIA										13BIB										14BIB										15BIB										16BIB										17BIB										18BIB																																																																																																																			
1	H	3	Li	4	Be	11	Na	12	Mg	19	K	20	Ca	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn	31	Ga	32	Ge	33	As	34	Se	35	Br	36	Kr	5	B	6	C	7	N	8	O	9	F	10	Ne	13	Al	14	Si	15	P	16	S	17	Cl	18	Ar	37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd	49	In	50	Sn	51	Sb	52	Te	53	I	54	Xe	55	Cs	56	Ba	57	La	72	Hf	73	Ta	74	W	75	Re	76	Os	77	Ir	78	Pt	79	Au	80	Hg	81	Tl	82	Pb	83	Bi	84	Po	85	At	86	Rn	87	Fr	88	Ra	89	Ac	104		105		106		107		108		109		110		111		112		113		114		115		116		117		118	



58	Ce	59	Pr	60	Nd	61	Pm	62	Sm	63	Eu	64	Gd	65	Tb	66	Dy	67	Ho	68	Er	69	Tm	70	Yb	71	Lu
90	Th	91	Pa	92	U	93	Np	94	Pu	95	Am	96	Cm	97	Bk	98	Cf	99	Es	100	Fm	101	Mn	102	Uu	103	Uub

Atomic weight is the average mass of the atoms of an element, taking into account the relative abundance of each isotope.

The number of the cell that is highlighted in red in the periodic table is the atomic number of the element.

Atomic weight is the average mass of the atoms of an element, taking into account the relative abundance of each isotope.

The number of the cell that is highlighted in red in the periodic table is the atomic number of the element.

This is a simplified representation of the periodic table. It does not show the relative positions of the elements in the periodic table.

Sargent-Welch
Scientific Supply

Sargent Welch Periodic Table

Lei Huang



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