



Student Exploration: Phase Changes

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: altitude, boil, boiling point, freeze, freezing point, gas, liquid, melt, melting point, phase, solid

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Marla wants a hard-boiled egg for breakfast. The recipe says boil water, then lower the heat so the water is simmering. But Marla is in a rush, so she keeps the heat on high as the egg cooks.

Do you think the egg will be done faster if the heat is kept on high as the water boils?

No

Explain your answer.

If the water is kept high the egg will cook inconsistently so the outer and inner layers will be at different stages.

Gizmo Warm-up

In the *Phase Changes* Gizmo, select **Micro view** and set the **Ice volume** to 50 cm³. Notice the nitrogen (), oxygen (), and water () molecules.

To start, create ice. Set **Add/remove heat energy** to -200 J/s, click **Play** (), and then click **Pause** () when the container is about half ice, half water. Set **Add/remove heat energy** back to 0 J/s, and then click **Play**.



1. Observe water molecules in the **solid** (ice), **liquid** (water), and **gas** (air) **phases**.

A. In which phase(s) do molecules move freely?

Gas

B. In which phase(s) are molecules held together?

Solid

C. What do you notice about the positions of the ice molecules?

They are all about equal distance apart and more uniform than the air or water molecules.

2. Click **Reset** (), and select **Macro view**. Set the **Ice volume** to 25 cm³.

A. In which phase(s) is water held in a defined shape?

Solid

B. In which phase(s) does the water take the shape of the container?

Liquid

Student Exploration Phase Changes Gizmo Answer Key

R Sandford



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Table of Contents Student Exploration Phase Changes Gizmo Answer Key

1. Understanding the eBook Student Exploration Phase Changes Gizmo Answer Key
 - The Rise of Digital Reading Student Exploration Phase Changes Gizmo Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Phase Changes Gizmo Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Phase Changes Gizmo Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Phase Changes Gizmo Answer Key
 - Personalized Recommendations
 - Student Exploration Phase Changes Gizmo Answer Key User Reviews and Ratings

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- Student Exploration Phase Changes Gizmo Answer Key and Bestseller Lists
- 5. Accessing Student Exploration Phase Changes Gizmo Answer Key Free and Paid eBooks
 - Student Exploration Phase Changes Gizmo Answer Key Public Domain eBooks
 - Student Exploration Phase Changes Gizmo Answer Key eBook Subscription Services
 - Student Exploration Phase Changes Gizmo Answer Key Budget-Friendly Options
- 6. Navigating Student Exploration Phase Changes Gizmo Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Student Exploration Phase Changes Gizmo Answer Key Compatibility with Devices
 - Student Exploration Phase Changes Gizmo Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Phase Changes Gizmo Answer Key
 - Highlighting and Note-Taking Student Exploration Phase Changes Gizmo Answer Key
 - Interactive Elements Student Exploration Phase Changes Gizmo Answer Key
- 8. Staying Engaged with Student Exploration Phase Changes Gizmo Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Phase Changes Gizmo Answer Key
- 9. Balancing eBooks and Physical Books Student Exploration Phase Changes Gizmo Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Phase Changes Gizmo Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Student Exploration Phase Changes Gizmo Answer Key
 - Setting Reading Goals Student Exploration Phase Changes Gizmo Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Student Exploration Phase Changes Gizmo Answer Key
 - Fact-Checking eBook Content of Student Exploration Phase Changes Gizmo Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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