

INORGANIC CHEMISTRY

451. Nitrogen can be obtained from air by removing
 (a) Oxygen (b) Hydrogen (c) Carbon dioxide (d) Both (a) and (c)
452. Which of the following carbide is a methanide?
 (a) Be_2C (b) CaC_2 (c) Al_2C_3 (d) Mg_2C_3
453. Caro's acid is
 (a) H_2SO_3 (b) $\text{H}_2\text{S}_2\text{O}_8$ (c) H_2SO_5 (d) $\text{H}_2\text{S}_2\text{O}_7$
454. Inorganic benzene is
 (a) $\text{B}_3\text{H}_3\text{N}_3$ (b) BH_3NH_3 (c) $\text{B}_3\text{H}_6\text{N}_3$ (d) $\text{H}_3\text{B}_3\text{N}_3$
455. For the reaction, $\text{P}(\text{White}) \longrightarrow \text{P}(\text{Red})$; $\Delta H = -18.4 \text{ kJ}$. It follows that
 (a) Red P is readily formed from white P
 (b) White P is readily formed from red P
 (c) White P can not be converted to red P
 (d) White P can be converted into red P and red P is more stable
456. Nitrogen dioxide cannot be obtained by heating
 (a) KNO_3 (b) $\text{Pb}(\text{NO}_3)_2$ (c) $\text{Cu}(\text{NO}_3)_2$ (d) AgNO_3
457. Reactivity of borazole is greater than that of benzene because:
 (a) Borazole is a non-polar compound
 (b) Borazole is a polar compound
 (c) Borazole is an electron deficient compound
 (d) Of localized electrons in it
458. When orthophosphoric acid is heated to 600°C , the product formed is
 (a) Phosphine, PH_3 (b) Phosphorus pentoxide, P_2O_5
 (c) Phosphorus acid, H_3PO_3 (d) Metaphosphoric acid, HPO_3
459. $\text{P}_4 + 3\text{NaOH} + 3\text{H}_2\text{O} \rightarrow \text{A} + 3\text{NaH}_2\text{PO}_2$, here, 'A' is
 (a) NH_3 (b) PH_3 (c) H_2PO_2 (d) H_3PO_4
460. SO_2 in the solid phase exists as
 (a) linear chain or cyclic trimer (b) planar triangular unit
 (c) square planar unit (d) none of these
461. Sulphur does not exist as S_2 molecule because
 (a) It is less electronegative (b) Does not form pπ-pπ bond
 (c) It has ability to exhibiting catenation (d) Has tendency to show variable oxidation state
462. Which of the following has S-S BOND?
 (a) $\text{H}_2\text{S}_2\text{O}_8$ (b) $\text{H}_2\text{S}_2\text{O}_7$ (c) $\text{H}_2\text{S}_2\text{O}_6$ (d) Mustard gas
463. Which of the following is correct order of stability order?
 (a) $\text{Pb}^{2+} > \text{Pb}^{4+}$ (b) $\text{Pb}^{4+} > \text{Pb}^{2+}$ (c) $\text{Si}^{2+} > \text{Si}^{4+}$ (d) $\text{Sn}^{4+} > \text{Sn}^{2+}$
464. Quartz is a crystalline variety of
 (a) Si (b) SiO_2 (c) Na_2SiO_3 (d) SiC
465. The equivalent weight of phosphoric acid in the reaction
 $\text{NaOH} + \text{H}_3\text{PO}_4 \longrightarrow \text{NaH}_2\text{PO}_4 + \text{H}_2\text{O}$
 (a) 25 (b) 49 (c) 59 (d) 98

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co₂ 10 h₂o 3 4 5 1 hbr 1 khco₃ 1 h₂o 1 kbr 1 co₂ 2 gabr₃ 3 na₂so₃ 1 ga₂ so₃ 3 6 nabr 3 sno 2 nf₃ 3 snf₂ 1 n₂o₃ using the following equation 2 naoh h₂so₄ 2 h₂o na₂so₄

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worth 10 points for problems 5 7 make sure to show your work using the factor label method please circle or highlight your answers

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