

**GO BELLY UP**

# Belly Up

**BM King**



**Belly Up:**

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web google classroom balance the following chemical equation  $\text{Mg(OH)}_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$  note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem  $7.4 \text{ x} \cdot \text{y} = \theta \pi$   $8.5 \cdot 2.0 = 9.6$

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web a common type of stoichiometric relationship is the mole ratio which relates the amounts in moles of any two substances in a chemical reaction we can write a mole ratio for a pair of substances by looking at the coefficients in front of each species in

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