

# Acids and Bases Worksheet 1

1. Finish the following reactions when acids are added to water.



2. What are the conjugate bases of these acids?

original acid	conjugate base
$\text{HNO}_3$	
$\text{H}_2\text{O}$	
$\text{H}_3\text{O}^+$	
$\text{H}_2\text{SO}_4$	
$\text{HBr}$	
$\text{HCO}_3^-$	

3. What are the conjugate acids of these bases?

original base	conjugate acid
$\text{OH}^-$	
$\text{H}_2\text{O}$	
$\text{HCO}_3^-$	
$\text{SO}_4^{2-}$	
$\text{ClO}_4^-$	

4. Which of the following represent conjugate acid-base pairs?



5. Calculate the  $[\text{H}^+]$  in a solution in which  $[\text{OH}^-] = 2.0 \times 10^{-2} \text{ M}$ . Is this solution acidic, neutral, or basic?

6. Calculate the  $[\text{OH}^-]$  in a solution in which  $[\text{H}^+] = 3.99 \times 10^{-5} \text{ M}$ . Is this solution acidic, neutral, or basic?

# Acids And Bases Study Guide Answer Key

**Arshad Iqbal**



**Acids And Bases Study Guide Answer Key:**

## **Acids And Bases Study Guide Answer Key** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Acids And Bases Study Guide Answer Key**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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