

১৫।  $x^2 - \sqrt{5}x + 1 = 0$  হলে  $x + \frac{1}{x}$  = কত?

- (ক)  $-2\sqrt{5}$  (খ)  $-\sqrt{5}$   
(গ)  $\sqrt{5}$  (ঘ)  $5\sqrt{5}$

১৬।  $x + y = 6$  এবং  $x - y = 4$  হলে  $5xy$  এর মান কত?

- (ক) 16 (খ) 24  
(গ) 25 (ঘ) 36

১৭।  $U = \{1, 2, 3, 4, 5, 6, 7\}$ ,  $A = \{3, 5, 6, 7\}$  হলে  $A$  সমান কত হবে?

- (ক)  $\{1, 2, 4\}$  (খ)  $\{2, 4, 5\}$   
(গ)  $\{2, 4, 6\}$  (ঘ)  $\{3, 5, 6, 7\}$

১৮।  $A = \{3, 4, 5\}$  এবং  $B = \{4, 5, 7\}$  হলে—

- i.  $A \cup B = \{3, 4, 5, 7\}$   
ii.  $A \cap B = \{4, 5\}$   
iii.  $A \subset B$

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) i ও iii  
(গ) ii ও iii (ঘ) i, ii ও iii

১৯।  $x^2 - x - 42$  এর উৎপাদকে বিশ্লেষিত রূপ নিচের কোনটি?

- (ক)  $(x-6)(x-7)$   
(খ)  $(x-6)(x+7)$   
(গ)  $(x+6)(x+7)$   
(ঘ)  $(x-7)(x+6)$

২০। কোনো বিন্দুটি  $x$ -অক্ষের উপর অবস্থিত?

- (ক)  $(0, -4)$  (খ)  $(-4, 0)$   
(গ)  $(0, 3)$  (ঘ)  $(3, -1)$

২১। কোন প্রকৃত ভগ্নাংশের লব ও হরের যোগফল 14 এবং বিয়োফল 8 হলে, ভগ্নাংশটি নিচের কোনটি?

- (ক)  $\frac{5}{9}$  (খ)  $\frac{9}{5}$   
(গ)  $\frac{3}{11}$  (ঘ)  $\frac{11}{3}$

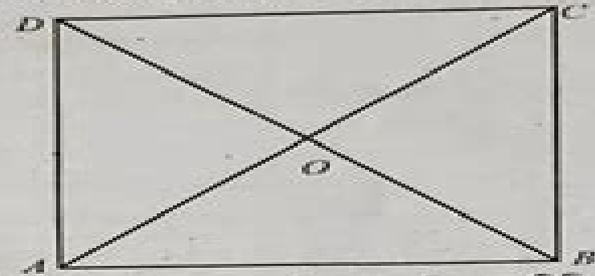
২২।  $m^2 - 6m - 1 = 0$  হলে  $m^2 + \frac{1}{m^2}$  = কত?

- (ক) 32  
(খ) 34  
(গ) 36  
(ঘ) 38

২৩। একটি সমকোণী ত্রিভুজের বাহুর ব্যাসার্ধ 4 সে.মি. এবং উচ্চতা 6 সে.মি.। ত্রিভুজটির বক্র পৃষ্ঠের ক্ষেত্রফল কত?

- (ক) 75.36 বর্গ সে.মি.  
(খ) 150.72 বর্গ সে.মি.  
(গ) 226.08 বর্গ সে.মি.  
(ঘ) 301.44 বর্গ সে.মি.

নিচের চিত্র অনুসারে ২৪ ও ২৫নং প্রশ্নের উত্তর দাও।



ABCD বর্জাক্ষের কর্ণের 'O' বিন্দুতে মিলিত হয়েছে এবং  $BC = 6$  cm.

২৪।  $\angle AOB$  = কত?

- (ক)  $30^\circ$  (খ)  $45^\circ$   
(গ)  $60^\circ$  (ঘ)  $90^\circ$

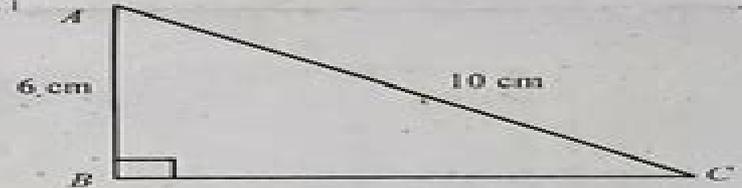
২৫। AC এর ঠিকানা কত?

- (ক) 6 সে.মি. (খ)  $6\sqrt{2}$  সে.মি.  
(গ)  $9\sqrt{2}$  সে.মি. (ঘ) 12 সে.মি.

২৬। যে চতুর্ভুজের দুই জোড়া সম্মিলিত বাহু সমান তাকে কী বলে?

- (ক) অয়ত (খ) বর্গ  
(গ) ঘূর্ণি (ঘ) সামান্তরিক

২৭।



ABC ত্রিভুজের ক্ষেত্রফল কত?

- (ক) 24 বর্গ সে.মি. (খ) 36 বর্গ সে.মি.  
(গ) 48 বর্গ সে.মি. (ঘ) 60 বর্গ সে.মি.

২৮। একটি ত্রিভুজের বাহুত্রয়ের অনুপাত  $x : x : x\sqrt{2}$  হলে, এর বৃহত্তম কোণটির মান কত?

- (ক)  $80^\circ$  (খ)  $85^\circ$   
(গ)  $90^\circ$  (ঘ)  $120^\circ$

২৯। চতুর্ভুজের চারটি কোণের সমষ্টি কত?

- (ক)  $360^\circ$  (খ)  $300^\circ$   
(গ)  $270^\circ$  (ঘ)  $180^\circ$

৩০। একটি ট্র্যাপিজিয়ামের সমান্তরাল বাহু দুটির ঠিকানা যথাক্রমে 13 সে.মি. এবং 8 সে.মি.। সমান্তরাল বাহুদ্বয়ের লম্ব দূরত্ব 4 সে.মি. হলে ট্র্যাপিজিয়ামটির ক্ষেত্রফল কত বর্গ?

- (ক) 42 বর্গ সে.মি. (খ) 52 বর্গ সে.মি.  
(গ) 84 বর্গ সে.মি. (ঘ) 104 বর্গ সে.মি.

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web mar 27 2023 a the lewis structure for water is represented by the given option a in this structure the oxygen atom is in the center and is bonded to two hydrogen atoms the oxygen atom has two lone pairs of electrons this structure follows the octet rule where each atom has a full valence shell with eight electrons

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web lewis structures represent the valence electrons present in the molecule as lone pairs unshared electrons drawn as dots or bonds shared electrons drawn as lines procedure for drawing lewis structures add up the total number of valence electrons that each atom contributes to the molecule polyatomic ion

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web the pcl<sub>3</sub> lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh<sub>3</sub> chlorine cl

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web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to each other with single bonds to form a skeleton structure

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and lone pairs surrounding the central atom and lastly identify the structure geometry

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