

# QUALITATIVE ANALYSIS (CATIONS)

Group	Group reagent	Cations (Basic radicals)	Composition & colour of precipitate
1	dil HCl	$Pb^{2+}$ (w)	$PbCl_2$ ; white
		$Ag^+$	$AgCl$ ; white (adv)
		$Hg_2^{2+}$	$Hg_2Cl_2$ ; white
2(A)	$H_2S$ in presence of dil HCl (insoluble in YAS)	$Pb^{2+}$ (w)	$PbS$ ; Black
		$Cu^{2+}$ (w)	$CuS$ ; Black
		$Hg^{2+}$	$HgS$ ; Black (adv)
		$Bi^{3+}$	$Bi_2S_3$ ; Black
		$Co^{2+}$ $Cd^{2+}$	$CoS$ ; yellow
2(B)	$H_2S$ in presence of dil HCl (soluble in YAS)	$As^{3+}, Sb^{3+}$	$As_2S_3$ ; yellow, $Sb_2S_3$ ; orange
		$Sn^{2+}, Sn^{4+}$	$SnS$ ; brown, $SnS_2$ ; yellow
3	$NH_4OH$ in presence of $NH_4Cl$	$Fe^{3+}$ (w)	$Fe(OH)_3$ ; red brown
		$Al^{3+}$ (w)	$Al(OH)_3$ ; gelatinous white
		$Cr^{3+}$	$Cr(OH)_3$ ; green (adv)
4	$H_2S$ in presence of $NH_4OH$ & $NH_4Cl$	$Zn^{2+}$ (w)	$ZnS$ ; white
		$Ni^{2+}$ (w)	$NiS$ ; black
5	$(NH_4)_2CO_3$ in presence of $NH_4OH$ <small>Mns (Bull/Pine)</small>	$Mn^{2+}$ (w)	$MnCO_3$ ; black, $MnS$ ; black
		$Ba^{2+}$ (w)	$BaCO_3$ ; white
		$Ca^{2+}$ (w)	$CaCO_3$ ; white
		$Sr^{2+}$	$SrCO_3$ ; white
6	$Na_2HPO_4$ in presence of $NH_4OH$	$Mg^{2+}$ (w)	$Mg(NH_4)PO_4$ ; white

# Qualitative Analysis Of Cations Lab Report Answers

**Tao Wei**



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