

### Revision worksheet , Grade 11, Chemistry.

1. What is isomerism? Explain functional isomerism, metamerism and tautomerism with the help of suitable examples.
2. What do you mean by resonance. Explain with the help of an example.
3. What do you mean by Electron Displacement Effect? Explain inductive effect with the help of a proper example.
4. Write a note on the following reactions:-
  - a. Sabatier and Senderens reaction
  - b. Friedel Craft's reaction
  - c. Isomerisation
  - d. Dehydrohalogenation
  - e. Markonikov's rule
  - f. Ozonolysis
  - g. Decarboxylation
  - h. Wurtz reaction
  - i. Kolbe's electrolysis
  - j. Reforming or aromatization
  - k. Dehalogenation
  - l. Peroxide effect
  - m. Reaction of an alkene with  $\text{KMnO}_4$
  - n. Dehydration
5. What is Grignard's reagent? How is it prepared? It was observed that Grignard's reagent should be stored in moisture proof containers. Why?
6. Identify the oxidizing agent, reducing agent, oxidant and reductant from the following using oxidation number method
$$\text{Zn} + 2\text{HCl} \longrightarrow \text{ZnCl}_2 + \text{H}_2$$
7. What is disproportionation reaction? Give an example.
8. What do you mean by an open system, closed system and an isolated system? Give an example.
9. What do you mean by saying that heat absorbed by a system is +ve and heat liberated by a system is -ve. Explain.
10. Explain the first law of Thermodynamics. Write its mathematical representation.
11. What do you mean by spontaneous and non spontaneous process. Give one example each.
12. State and explain Zeroth Law of Thermodynamics.
13. Explain Molarity, molality, mass percent and mole fraction.
14. Explain Aufbau principle, Hund's rule and Pauli's exclusion principle.
15. Draw the shapes of s,p and d orbitals.
16. What is ionization enthalpy? The ionization enthalpy of noble gases are very high. Why
17. What is electron gain enthalpy?

# Solution Chemistry Grade 11

**AW Rasmussen**



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## **Table of Contents Solution Chemistry Grade 11**

1. Understanding the eBook Solution Chemistry Grade 11
  - The Rise of Digital Reading Solution Chemistry Grade 11
  - Advantages of eBooks Over Traditional Books
2. Identifying Solution Chemistry Grade 11
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Solution Chemistry Grade 11
  - User-Friendly Interface
4. Exploring eBook Recommendations from Solution Chemistry Grade 11
  - Personalized Recommendations
  - Solution Chemistry Grade 11 User Reviews and Ratings
  - Solution Chemistry Grade 11 and Bestseller Lists
5. Accessing Solution Chemistry Grade 11 Free and Paid eBooks
  - Solution Chemistry Grade 11 Public Domain eBooks
  - Solution Chemistry Grade 11 eBook Subscription Services
  - Solution Chemistry Grade 11 Budget-Friendly Options

6. Navigating Solution Chemistry Grade 11 eBook Formats
  - ePub, PDF, MOBI, and More
  - Solution Chemistry Grade 11 Compatibility with Devices
  - Solution Chemistry Grade 11 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Solution Chemistry Grade 11
  - Highlighting and Note-Taking Solution Chemistry Grade 11
  - Interactive Elements Solution Chemistry Grade 11
8. Staying Engaged with Solution Chemistry Grade 11
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Solution Chemistry Grade 11
9. Balancing eBooks and Physical Books Solution Chemistry Grade 11
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Solution Chemistry Grade 11
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Solution Chemistry Grade 11
  - Setting Reading Goals Solution Chemistry Grade 11
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Solution Chemistry Grade 11
  - Fact-Checking eBook Content of Solution Chemistry Grade 11
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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