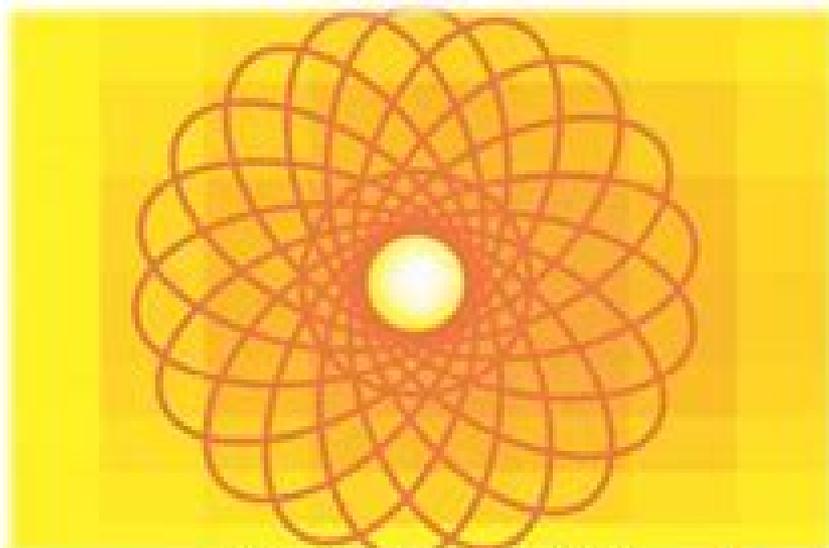


Unit 1: Historical Development of Scientific Theory

- Scientists conduct experiments, collect results, observe, and draw conclusions.
- They use this knowledge to develop general principles and theories.
- Theories describe observed phenomena and predict unobserved phenomena.
- New knowledge evolves these models.

- Science is not a list of facts but a way of thinking about the universe.
- New scientific knowledge, despite being based on experimental evidence, isn't always accepted.
- New knowledge requires sound experimental evidence and support from other scientists.



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Date	Person	What he did	Model of the atom	
440 BC	Leucippus	Philosopher	Matter is made of invisible units.	
400 BC	Democritus	Philosopher	Matter is made of small "atoms".	
384-322 BC	Aristotle	Philosopher	Matter is made of air, earth, fire and water.	
1704	Newton	Used the ideas of atoms to explain gas laws.	Atom is very small & solid with a mass.	
1803 (1811)	John Dalton (Avogadro)	<ul style="list-style-type: none"> - Dalton and others observed elements react in definite proportions by mass. - Avogadro's 1811 work contradicted Dalton's water formula. - Recognition of Avogadro's work took 50 years. 	<ul style="list-style-type: none"> - Atoms are solid spheres with measurable mass. - Elements are composed of same type atoms. - Compounds are composed of different type atoms. - Atoms rearranged during chemical reactions. 	

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