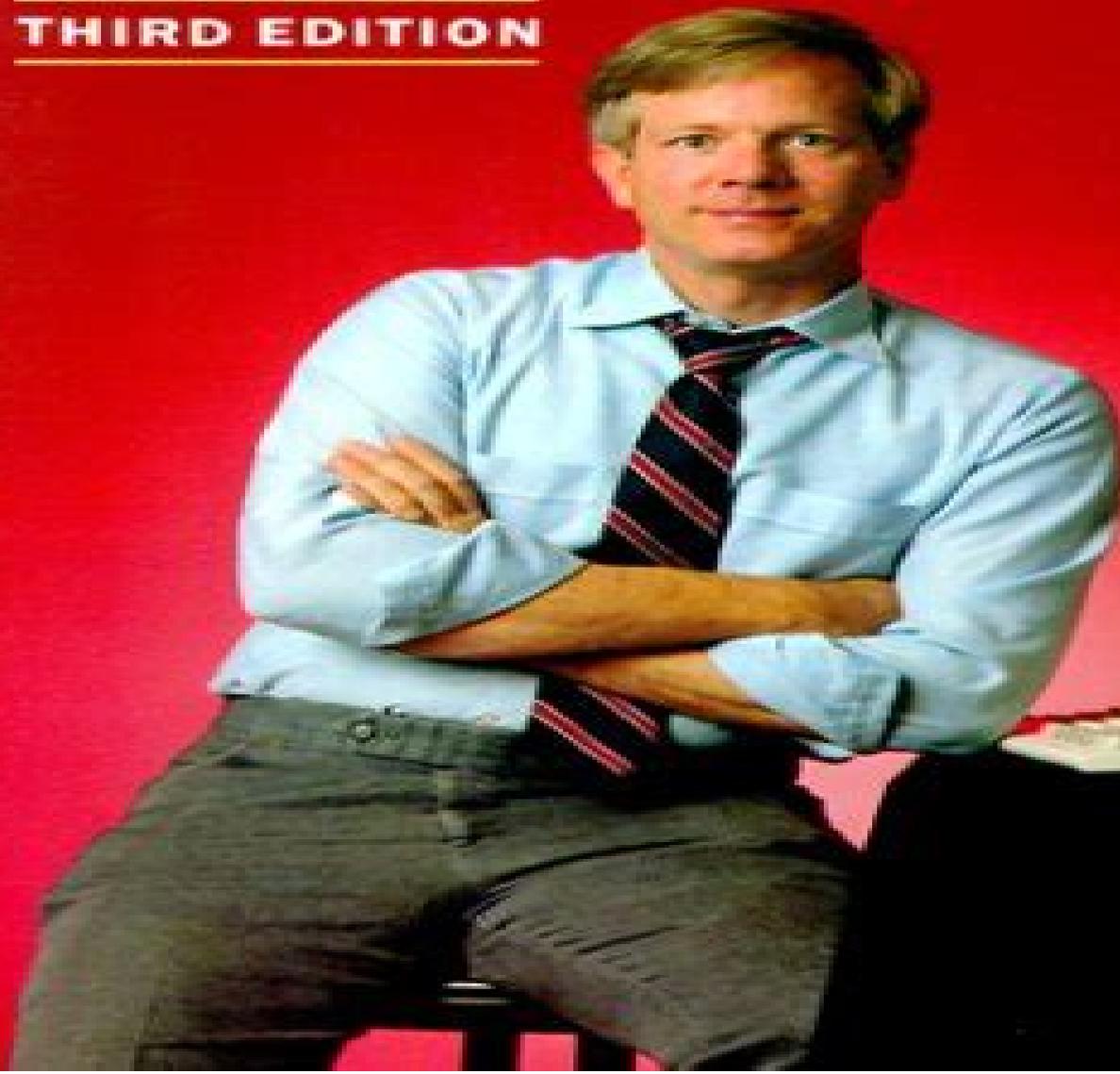


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$$m = \frac{M \cdot 1000}{1000d}$$

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$$M_{mix} = \frac{m_1 \cdot V_1 + m_2 \cdot V_2 + \dots + m_n \cdot V_n}{V_{total}}$$

1.5 x 480 + 1.2 x 520 + 480 + 520 + 720 + 624 + 1000 + 1344 = m_{q5} with increase in temperature which of these changes a molality b fraction of solute present in water

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