

Name _____ Date _____ Class _____

TEACHING TRANSPARENCY 16

Atomic Orbitals

Use with Chapter 5,
Section 5.2

1. What is the shape of an s orbital?

Spherical

2. What is the relationship between the size of an s orbital and the principal energy level in which it is found?

3. What is the shape of a p orbital? How many p orbitals are there in a sublevel?

4. How many electrons can each orbital hold?

2

5. Look at the diagrams of the p orbitals. What do x, y, and z refer to?

6. How many d orbitals are there in a given sublevel? How many total electrons can the d orbitals in a sublevel hold?

5 10

7. Which d orbitals have the same shape?

$x^2 - y^2$ and yz

8. What point in each diagram represents an atom's nucleus?

9. How likely is it that an electron occupying a p or a d orbital would be found very near an atom's nucleus? What part of the diagram supports your conclusion?

Teaching Transparency 16 Answers

Padhraic Smyth



Teaching Transparency 16 Answers:

This book delves into Teaching Transparency 16 Answers. Teaching Transparency 16 Answers is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Teaching Transparency 16 Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Teaching Transparency 16 Answers
 - Chapter 2: Essential Elements of Teaching Transparency 16 Answers
 - Chapter 3: Teaching Transparency 16 Answers in Everyday Life
 - Chapter 4: Teaching Transparency 16 Answers in Specific Contexts
 - Chapter 5: Conclusion
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