

Periodic Table and Trends

The Periodic Table: Elements are arranged on the periodic table based on their physical and chemical properties. It was Dmitri Mendeleev (1834-1907) who first organized a table listing the elements in order of atomic mass in vertical columns. Each of the columns ended when the chemical properties of the elements started to repeat themselves. Atoms are listed in increasing atomic number and generally in increasing atomic mass.

Part A

Using colours and/or shading, indicate the following:

- Label Hydrogen
- Location of the metals, metalloids, non-metals
- The boundary between the metals and non-metals
- The following families: alkali metals, alkaline earth metals, transition metals, halogens, noble gases, lanthanides (rare earth metals), and actinides
- Label the groups I to VIII

Be sure to include a key, if necessary.



Periodic Table Trends Assignment Answers

L Darling-Hammond



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