

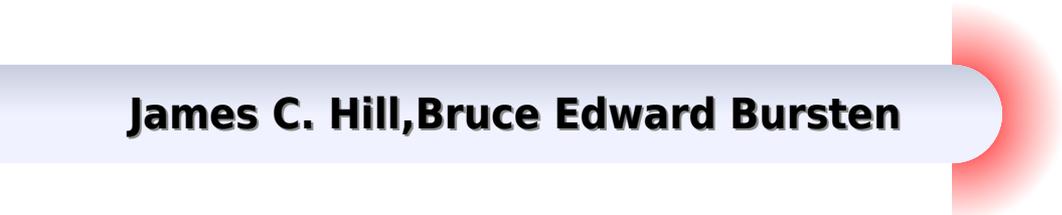
UNIT 5 COORDINATION COMPOUNDS

Multiple Choice Questions (MCQs):

- A purple compound 'M' has a formula $\text{Co}(\text{NH}_3)_6\text{Cl}_2$, on adding excess AgNO_3 to one mole of 'M' produces 2 moles of AgCl . If secondary valence of Co is 6 then correct formula of the complex will be?
 - $[\text{CoCl}_2(\text{NH}_3)_6]$
 - $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2$
 - $[\text{Co}(\text{NH}_3)_6\text{Cl}]\text{Cl}_2 \cdot \text{NH}_3$
 - $[\text{Co}(\text{NH}_3)_6\text{Cl}]\text{Cl}_2$
- Which among the following is not a double salt?
 - Mohr's Salt
 - Potash Alum
 - Carnallite
 - Potassium ferrocyanide.
- The correct order of the stoichiometries of AgCl formed when AgNO_3 in excess is treated with the complexes: $\text{CoCl}_2 \cdot 6\text{NH}_3$, $\text{CoCl}_2 \cdot 5\text{NH}_3$, $\text{CoCl}_2 \cdot 4\text{NH}_3$ respectively is (coordination valency is 6):
 - 3 AgCl , 1 AgCl , 2 AgCl
 - 3 AgCl , 2 AgCl , 1 AgCl
 - 2 AgCl , 3 AgCl , 1 AgCl
 - 1 AgCl , 3 AgCl , 2 AgCl
- Which among the following has highest molar conductivity?
 - $\text{K}_4[\text{Fe}(\text{CN})_6]$
 - $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2$
 - $[\text{Cu}(\text{NH}_3)_4]\text{SO}_4$
 - $[\text{Ni}(\text{CO})_4]$
- The primary and secondary valencies of $[\text{Fe}(\text{C}_2\text{O}_4)_3]^{3-}$ complexes are
 - 3 and 3
 - 3 and 6
 - 6 and 3
 - 0 and 3
- Coordination number of the central atom/ion is determined by
 - only by the number of pi bonds formed by the ligand with the central atom/ion.
 - both by the number of sigma and pi bonds formed by the ligand with the central atom/ion.
 - only by the number of sigma bonds formed by the ligand with the central atom/ion.
 - by the total number of electrons present in ligand.
- The correct IUPAC name of $\text{K}_3[\text{Cr}(\text{C}_2\text{O}_4)_3]$ is
 - Potassiumtrioxalatochromium(III)
 - Potassiumtrioxalatochromate(III)
 - Potassiumtrisoxalatochromate(III)
 - Potassium oxalatochromate(III)
- $[\text{Pt}(\text{NH}_3)_4\text{Cl}_2] \text{Br}_2$ complex can show :
 - Hydrated as well as ionization isomerism
 - Ionization as well as geometrical isomerism
 - Linkage as well as geometrical isomerism
 - Ionization as well as optical isomerism
- Possible isomerism in complexes $[\text{Co}(\text{NH}_3)_5(\text{NO}_2)]$ and $[\text{Co}(\text{NH}_3)_5(\text{NO}_2)]\text{Cl}_2$ respectively are:
 - Linkage and optical
 - Geometrical and linkage
 - Optical and ionization
 - Linkage and geometrical
- In the reactions $[\text{CoCl}_2(\text{NH}_3)_4]^{+} + \text{Cl}^{-} \rightarrow [\text{CoCl}_3(\text{NH}_3)_3] + \text{NH}_3$ two isomers of the product are obtained. The initial complex is
 - cis* isomer
 - trans* isomer
 - cis* or *trans* isomers
 - None of these

Coordination Compounds Multiple Choice Questions

James C. Hill, Bruce Edward Bursten



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