

	Reaction 1	Reaction 2	Reaction 3
1. Initial temperature	22.0°C	23.0°C	23.0°C
2. Final temperature	24.0°C	21.0°C	26.0°C
3. The change in temperature, ΔT	+2.0°C	-2.0°C	3.0°C
4. Heat, Q, absorbed by the solution = mass of soln. × ΔT × 1 cal/g °C Note: Mass = sum of masses of solids and liquids used.	151.6g × 2.0°C = 303.2 cal	160ml × 4.184 cal/g × (-2.0°C) = -640 cal	156.15g - 9.27g = 156.88g × 3.0°C = 470.64 cal
5. The amount of heat absorbed by the cup (neglect this quantity)	neglect	neglect	neglect
6. TOTAL amount of heat, Q, absorbed	303.2 cal	640 cal	470.64 cal
7. The number of moles of NaOH used in each reaction For solid: gNaOH × 1/MM NaOH For solution: L NaOH × Molarity	0.04 mol	0.04 mol	0.04 mol
8. The heat of reaction per mole of NaOH, ΔH. Include sign. (Line 6/Line 7)	7,580 cal	16,000 cal	11,766 cal

(FOR EACH)

Questions

- Write the net ionic equations for reactions 1, 2, and 3.
- In reaction 1, ΔH₁ represents the heat of solution of NaOH(s). Look at the net ionic equations for reactions 2 and 3 and make a statement concerning the significance of ΔH₂, that is, what has happened to produce the energy change. Also explain the meaning of ΔH₃.
- How does ΔH₂ compare with the sum of ΔH₁ + ΔH₃. Explain.
- Calculate the percent difference between ΔH₂ and the SUM of ΔH₁ + ΔH₃, using ΔH₂ as the reference.

HCl 23.0°C
NaOH 23.0°C

$$\frac{\text{Difference}}{\Delta H_2} \times 100\% =$$

_____ % difference

Answers To Lab 42 Neutralization Reactions

Tobias Bleicker



Answers To Lab 42 Neutralization Reactions:

This book delves into Answers To Lab 42 Neutralization Reactions. Answers To Lab 42 Neutralization Reactions is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Answers To Lab 42 Neutralization Reactions, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Answers To Lab 42 Neutralization Reactions
- Chapter 2: Essential Elements of Answers To Lab 42 Neutralization Reactions
- Chapter 3: Answers To Lab 42 Neutralization Reactions in Everyday Life
- Chapter 4: Answers To Lab 42 Neutralization Reactions in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Answers To Lab 42 Neutralization Reactions. This chapter will explore what Answers To Lab 42 Neutralization Reactions is, why Answers To Lab 42 Neutralization Reactions is vital, and how to effectively learn about Answers To Lab 42 Neutralization Reactions.

3. In chapter 2, this book will delve into the foundational concepts of Answers To Lab 42 Neutralization Reactions. The second chapter will elucidate the essential principles that must be understood to grasp Answers To Lab 42 Neutralization Reactions in its entirety.

4. In chapter 3, the author will examine the practical applications of Answers To Lab 42 Neutralization Reactions in daily life. The third chapter will showcase real-world examples of how Answers To Lab 42 Neutralization Reactions can be effectively utilized in everyday scenarios.

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6. In chapter 5, this book will draw a conclusion about Answers To Lab 42 Neutralization Reactions. This chapter will summarize the key points that have been discussed throughout the book.

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