

## 5-2 Review and Reinforcement Part II

### Quantum Theory and the Atom

Explain this principle, state the conditions of the phenomenon, and give an equation:

Photoelectric effect

Pauli exclusion principle

Spectroscopic evidence

Compton effect

Bohr's theory

Heisenberg

Principal energy levels

\_\_\_\_\_ called \_\_\_\_\_ 1. A model in which around the nucleus where electrons in shells orbit freely.

\_\_\_\_\_ electron theory \_\_\_\_\_ 2. The ground of an electron shell.

\_\_\_\_\_ quantum number \_\_\_\_\_ 3. Another development a principle energy level structure.

Pauli exclusion principle is states that no two electrons within an atom can have \_\_\_\_\_ and the same \_\_\_\_\_.

\_\_\_\_\_ quantum number spin \_\_\_\_\_.

\_\_\_\_\_ quantum number is called \_\_\_\_\_ 4. Spin, which progresses clockwise by increasing the \_\_\_\_\_ quantum number and \_\_\_\_\_.

\_\_\_\_\_ according to theory \_\_\_\_\_.

\_\_\_\_\_ principle energy level \_\_\_\_\_ 5. The level energy level structure.

\_\_\_\_\_ electron spin \_\_\_\_\_ 6. The structure of an atom, which is made of \_\_\_\_\_.

Compton effect is showing \_\_\_\_\_ when the \_\_\_\_\_ of the electron structure \_\_\_\_\_.

\_\_\_\_\_ 7. If an electron is used as a projectile when the probability of Compton scattering is \_\_\_\_\_.

- greater
- smaller
- higher energy
- constant

\_\_\_\_\_ 8. The low principal energy level of the hydrogen atom contains only one \_\_\_\_\_.

- electron
- proton
- neutron
- positron

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