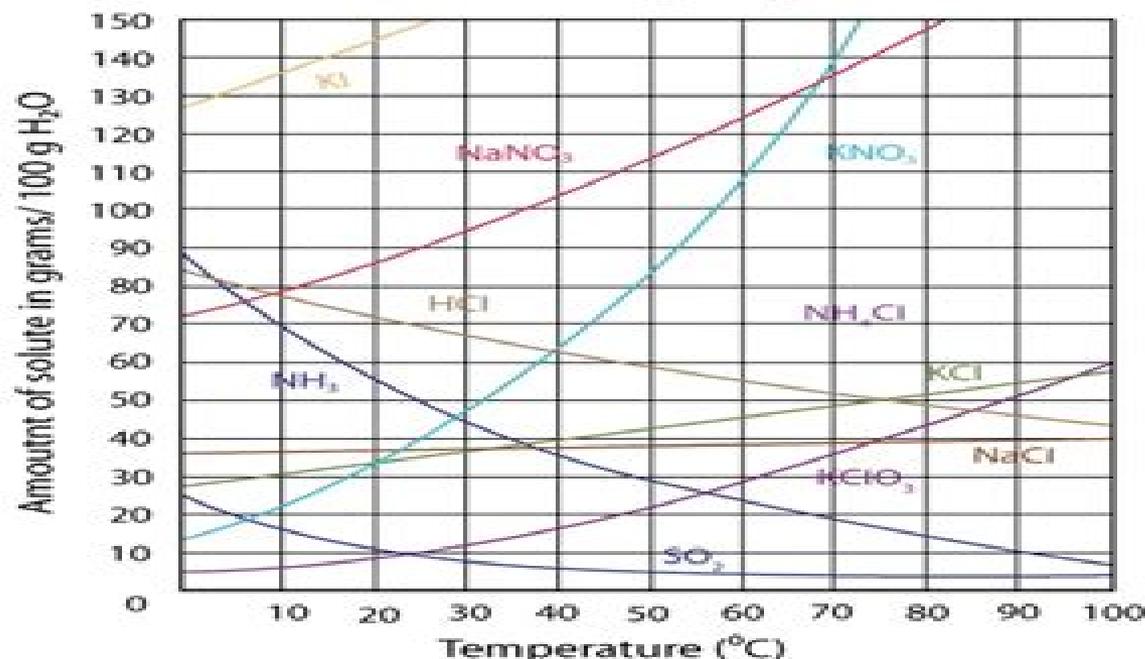




Interpreting Solubility Curve



1. Define solubility. _____
2. Using the solubility curve below, answer the given questions.



- I. What is the solvent in this solubility curve? _____
- II. Saturation points are given per how much of the solvent? _____
- III. What is the saturation point of NaNO₃ at 10°C, 40°C, & 80°C?
 10°C = _____ 40°C = _____ 80°C = _____
- IV. How much NH₃ can you dissolve in water at 10°C, 30°C, & 90°C?
 10°C = _____ 30°C = _____ 90°C = _____
- V. Which salt is the least soluble in water? _____
- VI. How many grams of potassium chloride can be dissolved in 200 g of water at 80 °C? _____
- VII. At 40 °C, how much potassium nitrate will be dissolved in 300 g of water? _____
- VIII. Which salt shows the least change in solubility from 0° to 100° C? _____
- IX. Which substance shows a decrease in solubility from 0° to 100° C? _____
- X. At 30 °C, 90 g of sodium nitrate is dissolved in 100 g water. Is this solution saturated, unsaturated, or supersaturated? _____

Solubility Temperature Graphs Chapter 14 Worksheet Answers

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Solubility Temperature Graphs Chapter 14 Worksheet Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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