# 6.1 ORGANIZING THE ELEMENTS

## Section Review

### Objectives
- Explain how elements are organized in a periodic table
- Compare early and modern periodic tables
- Identify three broad classes of elements

### Vocabulary
- periodic law
- metals
- nonmetals
- metalloids

## Part A  Completion

*Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.*

Chemists used the ___1___ of elements to sort them into groups.

The periodic table organizes the elements into vertical ___2___ and horizontal ___3___ in order of increasing ___4___. The table is constructed so that elements that have similar chemical properties are in the same ___5___. ___6___ have a high luster, or sheen, when cut. Most nonmetals are ___7___ at room temperature. Elements with properties that are similar to those of metals and nonmetals are called ___8___. Across the periodic table, the properties of elements become ___9___ metallic and ___10___ nonmetallic.

1. _____

2. _____

3. _____

4. _____

5. _____

6. _____

7. _____

8. _____

9. _____

10. _____

# Section 3 1 Organizing The Elements Answers

L Cohen

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