

Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

18 mol O₂ 1.

How many moles of oxygen are produced from twelve moles of Potassium chlorate?

21 mol O₂ 2.

How many moles of oxygen are produced from 14 moles of Potassium chlorate?

45 mol H₂O 3.

How many moles of water are produced from 22.5 moles of oxygen?

20 mol HCl 4.

Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.

21 mol Cl₂ 5.

Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



Chemistry Stoichiometry Problem Sheet 2 Answer Key

David Wilson



Chemistry Stoichiometry Problem Sheet 2 Answer Key:

The Enigmatic Realm of **Chemistry Stoichiometry Problem Sheet 2 Answer Key**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Chemistry Stoichiometry Problem Sheet 2 Answer Key** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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