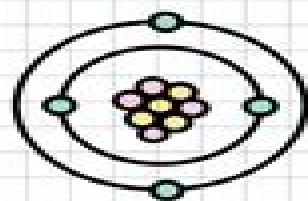


Atomic structure and the periodic table

Atomic structure



protons
neutrons
electrons

atoms have no overall charge

particle	relative mass	relative charge
proton	1	+1
neutron	1	0
electron	very small	-1

elements

elements - substances made up of atoms with the same atomic number

isotopes of an element - atoms with the same number of protons but different numbers of neutrons

relative atomic mass - the average mass number for an element

Equations, mixtures and chromatography

chemical formulas and equations

chemical formula - shows the proportion of atoms of each element in a compound

chemical equation - shows the overall change in a reaction

example:

reactants

products

word equation

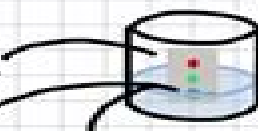
methane + oxygen → carbon dioxide + water

symbol equation



chromatogram

insoluble components stay on the base line

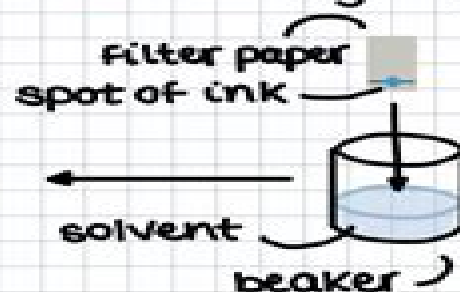


components separate as they move up the paper

submerge it



paper chromatography



filter paper

spot of ink

solvent

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Ted Lister, Janet Renshaw



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Table of Contents Chapter 7 Notes Atomic Structure And Periodicity

1. Understanding the eBook Chapter 7 Notes Atomic Structure And Periodicity
 - The Rise of Digital Reading Chapter 7 Notes Atomic Structure And Periodicity
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 7 Notes Atomic Structure And Periodicity
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 7 Notes Atomic Structure And Periodicity
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 7 Notes Atomic Structure And Periodicity
 - Personalized Recommendations
 - Chapter 7 Notes Atomic Structure And Periodicity User Reviews and Ratings
 - Chapter 7 Notes Atomic Structure And Periodicity and Bestseller Lists

5. Accessing Chapter 7 Notes Atomic Structure And Periodicity Free and Paid eBooks
 - Chapter 7 Notes Atomic Structure And Periodicity Public Domain eBooks
 - Chapter 7 Notes Atomic Structure And Periodicity eBook Subscription Services
 - Chapter 7 Notes Atomic Structure And Periodicity Budget-Friendly Options
6. Navigating Chapter 7 Notes Atomic Structure And Periodicity eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 7 Notes Atomic Structure And Periodicity Compatibility with Devices
 - Chapter 7 Notes Atomic Structure And Periodicity Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 7 Notes Atomic Structure And Periodicity
 - Highlighting and Note-Taking Chapter 7 Notes Atomic Structure And Periodicity
 - Interactive Elements Chapter 7 Notes Atomic Structure And Periodicity
8. Staying Engaged with Chapter 7 Notes Atomic Structure And Periodicity
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 7 Notes Atomic Structure And Periodicity
9. Balancing eBooks and Physical Books Chapter 7 Notes Atomic Structure And Periodicity
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 7 Notes Atomic Structure And Periodicity
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 7 Notes Atomic Structure And Periodicity
 - Setting Reading Goals Chapter 7 Notes Atomic Structure And Periodicity
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 7 Notes Atomic Structure And Periodicity
 - Fact-Checking eBook Content of Chapter 7 Notes Atomic Structure And Periodicity
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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