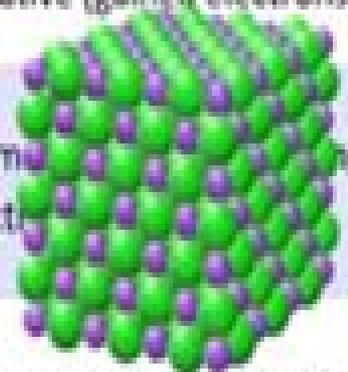


Atoms, Ions, & Molecules

	Atom	Ion	Molecule
Number of particles	 One 	One or more   	Two or more 
Charge	Neutral (equal protons and electrons)	Positive (lost electrons) negative (gained electrons)	Neutral (electrons shared between atoms)
Formation	Cannot be further broken down without losing chemical identity.	Formed by losing or gaining electrons. 	Formed by sharing electrons between atoms.
Examples	Helium (He), Oxygen (O), Iron (Fe)	Sodium ion (Na ⁺), Chloride ion (Cl ⁻), Sulfate ion (SO ₄ ²⁻)	Water (H ₂ O), Carbon dioxide (CO ₂), Sugar (C ₆ H ₁₂ O ₆)



Chapter 7 Compounds Ions And Molecules

JA Banks



Chapter 7 Compounds Ions And Molecules:

Reviewing **Chapter 7 Compounds Ions And Molecules**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Chapter 7 Compounds Ions And Molecules**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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