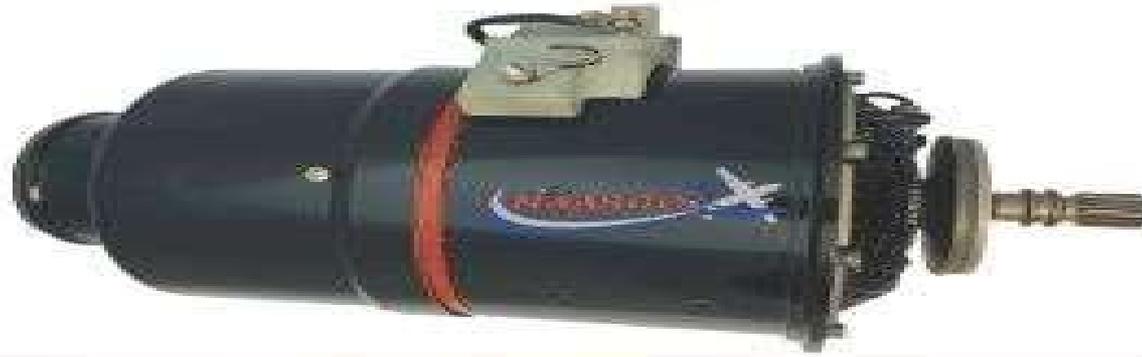


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# Apc Starter Generator Manual

**Lauren Gardner**



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chapter and WP title as well as all figures and tables contained within each WP Figures and tables are numbered sequentially within each WP The WP number is located at the top right of each page It is also located at the bottom of the page with the WP page number included 0001 1 would be page 1 of the General Information WP WP 0001 General Information Each WP starts on a right hand page This is done so you can remove a single WP from the paper TM if needed for a task Blank pages are assigned a number but it appears on the preceding or following page For example if page 0001 10 of a WP is blank page 0001 9 will have the number 0001 9 10 blank or if page 0001 1 of a WP is blank page 0001 2 will have the number 0001 1 blank 2 Each WP containing step by step maintenance or troubleshooting procedures will end with the words END OF TASK and each WP ends with the statement END OF WORK PACKAGE Think of each WP as a small standalone TM

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web answer a 1 explanation the entire reaction was multiplied by 2 therefore  $\Delta H$  must also be halved 3 answer c explanation option c is incorrect because 566 kJ/mol is the  $\Delta H$  per mole of  $\text{O}_2$  and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

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web q 140 g 4 184 j 16oc g oc q 9372 16 j 35 5 g x 1 mole kcn 0 5453 moles 65 1 g 0 5453 mole moles 5 when a 15 g sample of srcl<sub>2</sub> dissolves in 80 g of water in a calorimeter the temperature drops from 42 oc to 15 c calculate the  $\Delta h$  for the process  
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web thermochemistry answers and explanations 1 a a thermodynamic system has two means of exchanging energy with its surroundings heat flow and pressure volume work heat flow  $q$  is the transfer of thermal energy due to temperature difference work  $w$  is macroscopic energy transfer between the system and its surroundings that occurs

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web jan 10 2021  $\Delta h$  27 4 kJ mol and  $\Delta s$  29 8 j k sulfur trioxide so<sub>3</sub> is produced in large quantities in the industrial synthesis of sulfuric acid sulfur dioxide is converted to sulfur trioxide by reaction with oxygen gas write a balanced chemical equation for the reaction of so<sub>2</sub> with o<sub>2</sub> g and determine its  $\Delta g$