

Introduction

This course is a general chemistry course for students who are interested in pursuing a degree in any of the natural sciences and engineering disciplines. The course is designed to provide students with a solid foundation in chemistry, including the study of matter, energy, and the chemical and physical properties of substances. The course is divided into two semesters. The first semester covers the topics of matter, energy, and the chemical and physical properties of substances. The second semester covers the topics of acids and bases, and the chemical and physical properties of substances.

Prerequisites: High school chemistry or equivalent. [View the Prerequisite Chart for this course.](#)

Part 1: Matter and Energy



1. Matter and energy are the two fundamental concepts in chemistry. Matter is anything that has mass and occupies space. Energy is the capacity to do work. Matter and energy are interconvertible.
2. Matter is classified into pure substances and mixtures. Pure substances are further classified into elements and compounds. Mixtures are classified into homogeneous and heterogeneous mixtures.
3. The physical and chemical properties of matter are studied in detail.



10. Matter and energy are the two fundamental concepts in chemistry. Matter is anything that has mass and occupies space. Energy is the capacity to do work. Matter and energy are interconvertible.

Question	Answer	Correct Answer	Feedback	Marked Correct	Marked Correct
10.	Matter and energy are the two fundamental concepts in chemistry. Matter is anything that has mass and occupies space. Energy is the capacity to do work. Matter and energy are interconvertible.	Correct	10/10	10/10	10/10
11.	Matter and energy are the two fundamental concepts in chemistry. Matter is anything that has mass and occupies space. Energy is the capacity to do work. Matter and energy are interconvertible.	Correct	10/10	10/10	10/10
12.	Matter and energy are the two fundamental concepts in chemistry. Matter is anything that has mass and occupies space. Energy is the capacity to do work. Matter and energy are interconvertible.	Correct	10/10	10/10	10/10
13.	Matter and energy are the two fundamental concepts in chemistry. Matter is anything that has mass and occupies space. Energy is the capacity to do work. Matter and energy are interconvertible.	Correct	10/10	10/10	10/10

11. Matter and energy are the two fundamental concepts in chemistry. Matter is anything that has mass and occupies space. Energy is the capacity to do work. Matter and energy are interconvertible.

Part 2: Acids and Bases



14. Acids and bases are two important classes of compounds. Acids are substances that donate protons (H⁺) and bases are substances that accept protons (H⁺).
15. Acids and bases are interconvertible. The reaction between an acid and a base is called a neutralization reaction.



12. Complete the following table with the correct answer for each question. (10 marks)

Question	Answer	Correct Answer	Marked Correct	Marked Correct
12.	Acids and bases are two important classes of compounds. Acids are substances that donate protons (H ⁺) and bases are substances that accept protons (H ⁺).	Correct	10/10	10/10
13.	Acids and bases are interconvertible. The reaction between an acid and a base is called a neutralization reaction.	Correct	10/10	10/10
14.	Acids and bases are two important classes of compounds. Acids are substances that donate protons (H ⁺) and bases are substances that accept protons (H ⁺).	Correct	10/10	10/10
15.	Acids and bases are interconvertible. The reaction between an acid and a base is called a neutralization reaction.	Correct	10/10	10/10

13. Complete the following table with the correct answer for each question. (10 marks)

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