

**CHAPTER 4 PRACTICE TEST: SOLUTION STOICHIOMETRY**1) Which of the following are weak electrolytes? HCl,  $\text{HC}_2\text{H}_3\text{O}_2$ ,  $\text{NH}_3$ , KCl

A) HCl, KCl

B) HCl,  $\text{HC}_2\text{H}_3\text{O}_2$ ,  $\text{NH}_3$ , KClC)  $\text{HC}_2\text{H}_3\text{O}_2$ , KClD)  $\text{HC}_2\text{H}_3\text{O}_2$ ,  $\text{NH}_3$ E) HCl,  $\text{HC}_2\text{H}_3\text{O}_2$ , KCl*Answer: D Diff: 2*2) What are the spectator ions in the reaction between KOH (aq) and  $\text{HNO}_3$ (aq)?A)  $\text{K}^+$  and  $\text{H}^+$ C)  $\text{K}^+$  and  $\text{NO}_3^-$ E)  $\text{OH}^-$  onlyB)  $\text{H}^+$  and  $\text{OH}^-$ D)  $\text{H}^+$  and  $\text{NO}_3^-$ *Answer: C Diff: 2*3) The balanced net ionic equation for precipitation of  $\text{CaCO}_3$  when aqueous solutions of  $\text{Na}_2\text{CO}_3$  and  $\text{CaCl}_2$  are mixed is \_\_\_\_\_.A)  $2\text{Na}^+(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{Na}_2\text{CO}_3(\text{aq})$ 

E)

B)  $2\text{Na}^+(\text{aq}) + 2\text{Cl}^-(\text{aq}) \rightarrow 2\text{NaCl}(\text{aq})$  $\text{Na}_2\text{CO}_3(\text{aq}) + \text{CaCl}_2(\text{aq}) \rightarrow 2\text{NaCl}(\text{aq}) + \text{CaCO}_3(\text{s})$ C)  $\text{Na}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{NaCl}(\text{aq})$ D)  $\text{Ca}^{2+}(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{CaCO}_3(\text{s})$ *Answer: D Diff: 3*4) Which of the following are strong acids? HI,  $\text{HNO}_3$ , HF, HBr

A) HF, HBr

C) HI, HF, HBr

E) HI,  $\text{HNO}_3$ , HBrB) HI,  $\text{HNO}_3$ , HF, HBrD)  $\text{HNO}_3$ , HF, HBr*Answer: E Diff: 2*

5) A neutralization reaction between an acid and a metal hydroxide produces \_\_\_\_\_.

A) water and a salt

C) oxygen gas

E) ammonia

B) hydrogen gas

D) sodium hydroxide

*Answer: A Diff: 2*6) Which of these metals is the least easily oxidized? Na, Au, Fe, Ca, Ag

A) Na

B) Au

C) Fe

D) Ca

E) Ag

*Answer: B Diff: 3*

7) Based on the equations below, which metal is the most active?



A) Ni

B) Ag

C) Cu

D) Pb

E) N

*Answer: A Diff: 3*

8) What is the concentration (M) of KCl in a solution made by mixing 25.0 mL of 0.100 M KCl with 50.0 mL of 0.100 M KCl?

A) 0.100

B) 0.0500

C) 0.0333

D) 0.0250

E) 125

*Answer: A Diff: 3*

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**Anthony S. Fauci**



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web this quiz will test your knowledge on iv reconstitution calculation problems for dosage and calculation courses 1 the doctor s order says vancomycin 2 000 mcg iv the instructions on the vial of vancomycin says to reconstitute with sterile water for a concentration of 2mg ml then to dilute each 2 mg in 75 ml of sterile normal sale

**reconstitution problems flashcards quizlet** - Dec 12 2022

web oct 21 2023 answer dosage strength of 500 000 units ml you would administer 1 5 ml study with quizlet and memorize flashcards containing terms like single strength reconstitution calculations the physician orders 750 mg of cefazolin im q 8h he pharmacy sends the following vial of cefazolin

**drug dosage calculations nclex practice questions 100** - Aug 08 2022

web oct 3 2023 in this section are the practice problems and questions for nursing dosage calculations this nursing test bank set includes 100 questions included topics are dosage calculation metric conversions unit conversions parenteral medications and fluid input and output

5 10 reconstituted medication nursing skills 2e - Jul 19 2023

web practice problem reconstituted medication patient information name liam vang dob 04 04 20xx age 8 allergies nkda weight 60 kg provider order cefazolin 500 mg im every 8 hours medication supplied see figure 5 10 2 for the drug label of the medication as it is supplied figure 5 10 drug label of cefazolin

**module 4 reconstitution of solutions practice problems** - Feb 14 2023

web module 4 reconstitution of solutions practice problems answers using dimensional analysis problem dimensional analysis

order solu cortef 150 mg available a vial of solu cortef powder directions add 1 8ml of sterile water to yield a solution of 250mg/ml what is the order 150 mg

**reconstitution calculator** - Apr 04 2022

web jun 5 2023 the reconstitution calculator is a handy tool for all who need to prepare a solution using a dry ingredient and a diluent input the dosage which is the mass of the dry ingredient in the case of medications they come in vials and you can find the dosage on the label input the required or prescribed amount of the diluent which in case of

level 2 3 dosage calculation practice problems studocu - Sep 09 2022

web reconstitution dosage calculation practice problems 1 order 0 gram augmentin po q 8 hours how many ml will you give per dose order cefizox 0 g im q12h available cefizox 1 g directions for mixing add 3 ml sterile water or ns for injection to make 3 ml 270 mg per ml how many ml will you give per dose