

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON_METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

<p>1.) H₂ hydrogen is diatomic:</p> $\text{H} + \text{H} \longrightarrow \text{H} \cdot \cdot \text{H} \longrightarrow \text{H}-\text{H} \quad \text{Single Bond}$
<p>2.) F₂ fluorine is diatomic:</p> $\text{F} + \text{F} \longrightarrow \cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot \longrightarrow$
<p>3.) O₂ oxygen is diatomic:</p> $\text{O} + \text{O} \longrightarrow \quad \quad \quad \longrightarrow \text{O}=\text{O} \quad \text{Double Bond}$
<p>4.) N₂ nitrogen is diatomic. Is this a triple bond?</p> $\text{N} + \text{N}$
<p>5.) BF₃ you need 3 fluorine atoms here:</p> $\text{B} + \text{F}$
<p>6.) Ammonia NH₃ hint: how many hydrogen atoms are needed?</p> $\text{N} + \text{H}$
<p>7.) Carbon dioxide CO₂</p> $\text{O} + \text{C} + \text{O}$
<p>8.) Methane CH₄ careful here 4 hydrogen atoms needed</p> $\text{C} + \text{H}$
<p>9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!</p> $\text{H} + \text{O} + \text{H}$
<p>10.) SO₂ hint: one pair of electrons from sulfur must be slit up for this one to work.</p> $\text{O} + \text{S} + \text{O}$

Chapter 8 Covalent Bonding Worksheet Answers

Fruitypiore

JA Banks



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