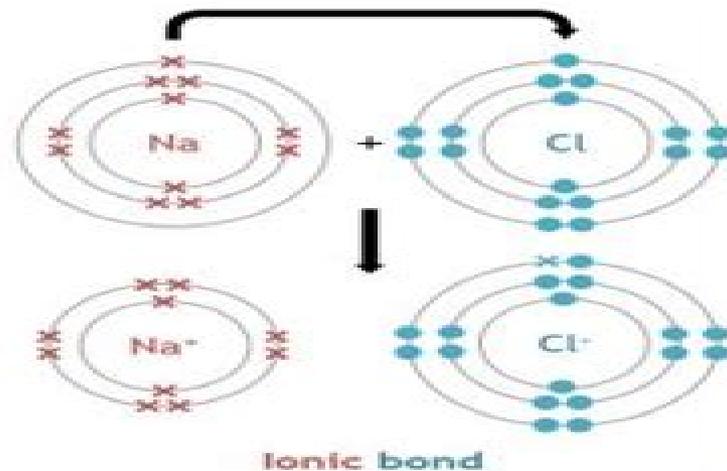


Ionic bonding revision (KS4)

Questions

1. Use the diagrams to describe how sodium and chlorine atoms react to form sodium chloride. (6 marks)



2. Complete the table using the charges on the ions to deduce the formula of each ionic compound.

(5 marks)

Metal	Charge on metal ion	Non-metal	Charge on non-metal ion	Name of ionic compound	Formula of ionic compound
Calcium		Bromine	1-	Calcium bromide	
Magnesium	2+	Sulfur			MgS
		Oxygen	2-	Lithium oxide	Li <sub>2</sub> O
Potassium	1+		1-		KI
Magnesium	2+	Chlorine		Magnesium chloride	

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