

AP Chemistry: Bonding Multiple Choice

41. Which of the following molecules has the shortest bond length?
(A) N₂ (B) O₂ (C) Cl₂ (D) Br₂ (E) I₂

51. Pi bonding occurs in each of the following species EXCEPT...
(A) CO₂ (B) C₂H₄ (C) CN⁻ (D) C₆H₆ (E) CH₄

60. Which of the following has a zero dipole moment?
(A) HCN (B) NH₃ (C) SO₂ (D) NO₂ (E) PF₅

8. Use the following answers for questions 8 - 9.

- (A) A network solid with covalent bonding
- (B) A molecular solid with zero dipole moment
- (C) A molecular solid with hydrogen bonding
- (D) An ionic solid
- (E) A metallic solid

8. Solid ethyl alcohol, C₂H₅OH

9. Silicon dioxide, SiO₂

80. For which of the following molecules are resonance structures necessary to describe the bonding satisfactorily?
(A) H₂S (B) SO₂ (C) CO₂ (D) OF₂ (E) PF₃

Hydrogen Halide	Normal Boiling Point, °C
HF	19
HCl	-85
HBr	-67
HI	-35

18. The liquefied hydrogen halides have the normal boiling points given above. The relatively high boiling point of HF can be correctly explained by which of the following?

- (A) HF gas is more ideal.
- (B) HF is the strongest acid.
- (C) HF molecules have a smaller dipole moment.
- (D) HF is much less soluble in water.
- (E) HF molecules tend to form hydrogen bonds.

42. The SbCl₅ molecule has trigonal bipyramid structure. Therefore, the hybridization of Sb orbitals should be...
(A) sp² (B) sp³ (C) sp²d (D) sp³d (E) sp³d²

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