

PERIODIC TABLE OF THE ELEMENTS

Table of Selected Radioactive Isotopes

GROUP 1IA		3IIIA										12IIIB										18VIII
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18					
1 H		3 Li	4 Be									5 B	6 C	7 N	8 O	9 F	10 Ne					
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar					
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr					
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe					
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn					
87 Fr	88 Ra	89 Ac	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118					



58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

Atomic number is the number of protons in the nucleus of the atom and is the defining characteristic of an element.

Atomic weight is the average mass of the atoms of an element, taking into account the relative abundance of each isotope.

The color of the element indicates its classification: alkali metal, alkaline earth metal, transition metal, main group element, lanthanide, actinide, noble gas, and halogen.

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