



Introduction to working as a Chemist.

Use a highlighter as you work through this booklet. ESL students, make vocab lists of the terminology that is defined or described here.

Assessment Statements

- | | |
|--|---|
| <ul style="list-style-type: none"> <input type="checkbox"/> Outline the Scientific Method <input type="checkbox"/> Identify independent, dependent & controlled variables. <input type="checkbox"/> Demonstrate correct and safe use of laboratory equipment. <input type="checkbox"/> Distinguish between quantitative and qualitative data <ul style="list-style-type: none"> > <i>Qualitative is descriptive, not using numbers, e.g. properties, colour, gas production, sound, odor.</i> > <i>Quantitative is mathematical, including</i> <ul style="list-style-type: none"> ↳ <i>Raw data (as recorded)</i> ↳ <i>Processed (after calculations)</i> <input type="checkbox"/> Distinguish between accuracy and precision <ul style="list-style-type: none"> > <i>Accuracy describes the correctness of a set of measurements</i> > <i>Precision describes the repeatability of a set of measurements</i> | <ul style="list-style-type: none"> <input type="checkbox"/> State SI units of measurement for <ul style="list-style-type: none"> > <i>Mass: grams (g)</i> > <i>Length: Metres (m), millimeters (mm)</i> > <i>Concentration: molarity (M)</i> > <i>Temperature: Degrees Celsius (°C)</i> > <i>Volume: Litres (L) or millilitres (mL)</i> > <i>Density: grams per cubic centimeter (g/cm³)</i> > <i>Conductivity: Siemens (S)</i> > <i>Acidity or alkalinity (pH)</i> <input type="checkbox"/> Energy: Joules (J or kJ) Determine the uncertainty of digital and analog measuring tools <ul style="list-style-type: none"> > <i>Usually ± 1 of the smallest division</i> <input type="checkbox"/> Calculate the % error of a given or recorded set of values <ul style="list-style-type: none"> > $\frac{(Measured - True)}{True} \times 100$ <input type="checkbox"/> State values to appropriate numbers of significant figures <input type="checkbox"/> Calculate values to appropriate numbers of significant figures |
|--|---|

Related Concepts:

- 01.1 Element symbols (Quia quiz: <http://www.quia.com/quiz/3246640.htm>)
- 01.2 Accuracy, Precision, Error, Uncertainty (Quia quiz: <http://www.quia.com/quiz/3507626.htm>)
- 01.3 SigDigs and Notation (Quia quiz: <http://www.quia.com/quiz/3504378.htm>)

Command Terms: Be sure you understand what is being asked of you

Analyse Break down to identify key parts. Interpret information to reach conclusions.

Apply	Use knowledge & understanding in response to a real situation or given circumstances.
Calculate	Obtain a numerical answer, showing relevant stages in the working.
Define	Give the precise meaning of a word, phrase, concept or physical quantity.
Describe	Give a detailed account or picture of an event, pattern, process or pattern.
Design	Produce a plan, simulation or model.
Determine	Obtain the only possible answer.
Discuss	Offer a considered and balanced review that includes a range of arguments, factors or hypotheses. Opinions or conclusions should be presented clearly and supported by appropriate evidence.
Distinguish	Make clear the differences between two or more concepts or items.
Evaluate	Assess the implications and limitations; make judgments about the ideas, works, solutions or ideas in relation to selected criteria.
Explain	Give a detailed account including reasons or causes.
Identify	Provide an answers from a number of possibilities. Recognise and state briefly a distinguishing fact or feature.
Label	Add title, labels or brief explanation(s) to a diagram or graph.
List	Give a sequence of answers with no explanation.
Outline	Give a brief account.
Recall	Remember or recognize from prior learning experiences.
State	Give a specific name, value or other brief answer without explanation or calculation.
Suggest/Propose	Propose a solution, hypothesis or other possible answer.

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Roman Wölfel



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most current information is included *Introduction to General, Organic, and Biochemistry* Morris Hein,1997 The authors
recognize that both science and mathematics may be daunting subjects for many students taking this course With this in

mind they have anticipated where students might stumble and have paced and organized this text to help them through Their goal is to make the material interesting and relevant so students understand the basic chemical principles related to their career The authors emphasize problem solving and provide a range of practice exercises As in previous editions the text first presents the basic concepts of general chemistry and then moves into organic and biochemistry In this edition the first two sections have been revised primarily to improve explanations and include new pedagogical features The biochemistry portion has been thoroughly updated to include coverage of many recent developments and emerging technologies in the field

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