

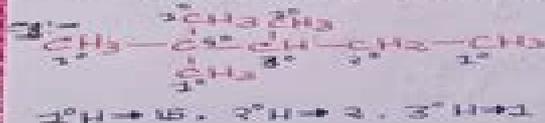
* Degree of Carbon :-

- 1° Carbon on. Primary carbon
- 2° Carbon on. Secondary Carbon
- 3° Carbon on. Tertiary Carbon
- 4° Carbon on. Quaternary Carbon



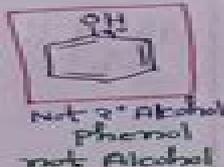
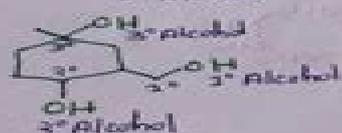
* Degree of Hydrogen :-

same as degree of carbon.



* Degree of Alcohol :-

same as degree of carbon to which it is attached.



* Degree of Halogen :-

same as degree of carbon to which it attached

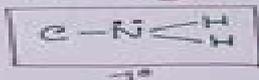


1°Cl → 1, 2°OH → 2, 3°OH → 3



* Degree of Nitrogen/ Amine :-

1°/ Primary nitrogen/ Amine → Nitrogen attached with carbon from one side



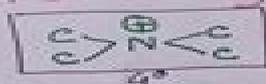
2°/ Secondary nitrogen/ Amine → Nitrogen attached with carbon from 2 side



3°/ Tertiary nitrogen/ Amine → Nitrogen attached with carbon from 3 side



4°/ quaternary nitrogen/ Amine → Nitrogen attached with carbon from 4 sides and holds one +ve charge



* Degree of Unsaturation :- (DU)

Double bond Equivalent / Index of Hydrogen Deficiency

$$DU = \frac{\text{No. of H in corresponding Alkane} - \text{No. of H in given compound}}{2}$$



1 double bond = 1 hydrogen deficiency
 or. 1 ring = [DU = 1]
 2 double bond = 2 H- deficiency or [DU = 2]
 or. 2 ring or. 2 triple bond

Chemistry S Notes

Sebastian Brünink



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Table of Contents Chemistry S Notes

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 - The Rise of Digital Reading Chemistry S Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry S Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry S Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry S Notes
 - Personalized Recommendations

- Chemistry S Notes User Reviews and Ratings
- Chemistry S Notes and Bestseller Lists
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 - Highlighting and Note-Taking Chemistry S Notes
 - Interactive Elements Chemistry S Notes
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry S Notes
- 9. Balancing eBooks and Physical Books Chemistry S Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry S Notes
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry S Notes
 - Setting Reading Goals Chemistry S Notes
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry S Notes
 - Fact-Checking eBook Content of Chemistry S Notes

- Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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