

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON_METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H ₂ hydrogen is diatomic: $\text{H} + \text{H} \longrightarrow \text{H} \cdot \cdot \text{H} \longrightarrow \text{H}-\text{H} \quad \text{Single Bond}$
2.) F ₂ fluorine is diatomic: $\text{F} + \text{F} \longrightarrow \cdot \cdot \cdot \cdot \text{F} \cdot \cdot \cdot \cdot \text{F} \cdot \cdot \cdot \cdot \longrightarrow$
3.) O ₂ oxygen is diatomic: $\text{O} + \text{O} \longrightarrow \quad \quad \quad \longrightarrow \text{O}=\text{O} \quad \text{Double Bond}$
4.) N ₂ nitrogen is diatomic. Is this a triple bond? $\text{N} + \text{N}$
5.) BF ₃ you need 3 fluorine atoms here: $\text{B} + \text{F}$
6.) Ammonia NH ₃ hint: how many hydrogen atoms are needed? $\text{N} + \text{H}$
7.) Carbon dioxide CO ₂ $\text{O} + \text{C} + \text{O}$
8.) Methane CH ₄ careful here 4 hydrogen atoms needed $\text{C} + \text{H}$
9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!! $\text{H} + \text{O} + \text{H}$
10.) SO ₂ hint: one pair of electrons from sulfur must be slit up for this one to work. $\text{O} + \text{S} + \text{O}$

Chapter 8 Covalent Bonding Worksheet Answers

David Baud



Chapter 8 Covalent Bonding Worksheet Answers:

Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Chapter 8 Covalent Bonding Worksheet Answers**

In a global inundated with monitors and the cacophony of quick connection, the profound energy and emotional resonance of verbal art frequently disappear into obscurity, eclipsed by the constant onslaught of noise and distractions. However, nestled within the lyrical pages of **Chapter 8 Covalent Bonding Worksheet Answers**, a interesting perform of literary elegance that impulses with fresh feelings, lies an unique journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, that enchanting opus manuals visitors on a psychological odyssey, delicately revealing the latent possible and profound impact stuck within the complex internet of language. Within the heart-wrenching expanse of this evocative examination, we shall embark upon an introspective exploration of the book is key styles, dissect their fascinating writing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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