

- A) 4
- B) 1
- C) 0
- D) 2
- E) 3

38) In which of the molecules below is the carbon-carbon distance the shortest?

- A)  $\text{H}_2\text{C} = \text{CH}_2$
- B)  $\text{H} - \text{C} \equiv \text{C} - \text{H}$
- C)  $\text{H}_3\text{C} - \text{CH}_3$
- D)  $\text{H}_2\text{C} = \text{C} = \text{CH}_2$
- E)  $\text{H}_3\text{C} - \text{CH}_2 - \text{CH}_3$

39) Of the bonds  $\text{C} - \text{N}$ ,  $\text{C} = \text{N}$ ,  $\text{C} \equiv \text{N}$  the  $\text{C} - \text{N}$  bond is \_\_\_\_\_.

- A) strongest/shortest
- B) strongest/longest
- C) weakest/shortest
- D) weakest/longest
- E) intermediate in both strength and length

40) Of the possible bonds between carbon atoms (single, double, and triple), \_\_\_\_\_.

- A) a triple (sigma + 2pi) bond is longer than a single bond
- B) a double (sigma + pi) bond is stronger than a triple bond
- C) a single (sigma) bond is stronger than a triple bond
- D) a double (sigma + pi) bond is longer than a triple bond
- E) a single (sigma) bond is stronger than a double bond

41) The ion  $\text{ICl}_4^-$  has \_\_\_\_\_ valence electrons.

- A) 34
- B) 35
- C) 36
- D) 28
- E) 8

42) The ion  $\text{NO}^-$  has \_\_\_\_\_ valence electrons.

- A) 15
- B) 14
- C) 16
- D) 10
- E) 12

43) The Lewis structure of  $\text{AsH}_3$  shows \_\_\_\_\_ nonbonding electron pair(s) on As.

- A) 0
- B) 1
- C) 2
- D) 3
- E) This cannot be determined from the data given.

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