

Physical Science Chapter 6 Test Review Sheet 2014-15

Assessments- This is the portion of the test which you, the student, will be permitted to retest on.

On the assessment, students will be required to:

1. Write out the names and formulas for a binary ionic compound.
2. Write out the names and formulas for a covalent (molecular) compound.

Students will know the following prefixes used for naming covalent/molecular compounds:

Mono-	1	Di-	2	Tri-	3
Tetra-	4	Penta-	5	Hexa-	6
Hepta-	7	Octa-	8	Nona-	9
Deca-	10				

Multiple Choice/Short Answer/Essay

1. Students will know, understand, and be able to apply the following terms:

electron dot diagram	ion	anion	cation
chemical bond	ionic bond	chemical formula	
covalent bond	molecule	polar covalent bond	metallic bond
polyatomic ion	semiconductor	molecular formula	empirical formula
hydrogen bond	formula unit	nonpolar covalent bond	

2. Students will know the composition, physical and chemical properties of ionic and covalent compounds.
3. Students will be able to predict and explain why elements lose or gain electrons to form ions.
4. Students will describe which groups in the periodic table are most willing to lose or gain electrons and explain why.
5. Students will know the different diatomic molecules as described in class (incomplete list in textbook)
6. Students will know what factors play a role in whether a molecular compound will be polar or nonpolar.

Physical Science Chapter 6 Test

RJ Alexander



Physical Science Chapter 6 Test:

Reviewing **Physical Science Chapter 6 Test**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Physical Science Chapter 6 Test**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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the different types of transport mechanisms across cell membranes are as follows simple diffusion facilitated diffusion

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transport ions from high concentration to low concentration transport ions such as sodium na potassium k calcium ca 2 and chloride cl because ions are not soluble in

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