

NAME \_\_\_\_\_

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## COUNTING BY MEASURING MASS

**Purpose:** To determine the mass of several samples of chemical compounds and use the data to count atoms.

**Materials:** safety goggles, tablespoon measure, weighing paper, small beaker or cup, scale, water, piece of aluminum, table salt (NaCl), baking soda (NaHCO<sub>3</sub>).

- Procedure:**
1. Measure the mass of about 50 mL of water.
  2. Measure the mass of the aluminum bar or a large ball of aluminum foil.
  3. Measure the mass of one tablespoon of NaCl.
  4. Measure the mass of one tablespoon of NaHCO<sub>3</sub>.
  5. Complete the table below showing all calculations.

**Data Table** (Show your work on a separate sheet of paper):

|                        | H <sub>2</sub> O (l) | Al (s) | NaCl (s) | NaHCO <sub>3</sub> (s) |
|------------------------|----------------------|--------|----------|------------------------|
| Mass (grams)           |                      |        |          |                        |
| Molar mass (g/mol)     |                      |        |          |                        |
| Moles of each compound |                      |        |          |                        |
| Moles of each element  |                      |        |          |                        |
| Atoms of each element  |                      |        |          |                        |

# Counting By Measuring Mass Lab Answer Key Xeocigsore

**D Keegan**



**Counting By Measuring Mass Lab Answer Key Xeocigore:**

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