

4. pH of Prepared Buffer Solutions

Buffer Solution	Acetic Acid; 4.5%, mL	Sodium Acetate Solution, mL	Recorded pH
A	5 mL	5 mL	4.7
B	5 mL	1 mL	3.9
C	10 mL	1 mL	3.7
D	1 mL	10 mL	5.7
E	1 mL	5 mL	5.3

III. Calculations/Results

*show all calculations with units

1. Sodium acetate moles _____
(molar mass = 82.0 g/mol)
2. Molarity of sodium acetate = _____
3. Concentration of acetic acid (from bottle) = _____
4. Molarity of acetic acid = _____
(molar mass = 60.0 g/mol; density of the solution is 1.0 g/mL)

Buffer Solution Lab Report

Guillaume Favre



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