

## Acids, Bases, Buffers and pH Data Tables

### Part A.

	Solution	pH Meter		pH paper		Universal Indicator	
		Reading	$[H_3O^+]$	color	pH value	color	pH value
	Control ( $H_2O$ )						
a.	0.1 M $NH_3$	11.00	$10^{-11} = 1^{-11}$	blue	12	purple	10
b.	0.1 M NaOH	12.85	$10^{-12.85} = 1.41^{-13}$	blue	13	purple	10
c.	0.1 M $H_3BO_3$	6.5	$10^{-6.5} = 3.16^{-7}$	orange yellow	4.5	yellow	6.5
d.	0.1 M HCl	1.40	$10^{-1.40} = .039$	orange	3	red/pink	3

### Part B.

	Solution	pH Meter		Universal Indicator	
		Reading	$[H_3O^+]$	color	pH value
	Control ( $H_2O$ )				
a.	0.1 M $Na_2CO_3$	11.42	$10^{-11.42} = 3.80^{-12}$	purple	10
b.	0.1 M $(NH_4)_2SO_4$	6.58	$10^{-6.58} = 2.63^{-7}$	yellow	6
c.	0.1 M NaCl	6.80	$10^{-6.80} = 1.58^{-7}$	bright green	6.5

### Part C.

	Solution	pH Meter
		Reading
	Control ( $H_2O$ )	
a.	Acetic Acid	3.05
b.	Sodium Acetate	7.15
c.	Mixture of a & b	4.83

# Ph Properties Of Buffer Solutions Pre Lab Answers

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**Ph Properties Of Buffer Solutions Pre Lab Answers:**

## Adopting the Track of Appearance: An Mental Symphony within **Ph Properties Of Buffer Solutions Pre Lab Answers**

In a global eaten by screens and the ceaseless chatter of quick conversation, the melodic splendor and psychological symphony produced by the written word frequently diminish into the background, eclipsed by the persistent noise and disturbances that permeate our lives. However, nestled within the pages of **Ph Properties Of Buffer Solutions Pre Lab Answers** a marvelous literary treasure full of organic emotions, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, this captivating masterpiece conducts viewers on a mental journey, well unraveling the concealed tunes and profound impact resonating within each cautiously constructed phrase. Within the depths with this moving review, we can investigate the book is main harmonies, analyze its enthralling writing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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