

Unit 1: Atomic Structure

(Also located on _____)

D. Ways to describe atomic info:

1. _____
" _____

Example: An atom has an atomic number of 8. How many protons are present in this atom?

1. _____
" _____
" _____

You Try:

Atoms of element Hydrogen have: _____

Atoms of element Oxygen have: _____

Atoms of element Iron have: _____

1. _____
" _____
" _____

Example: An atom has 6 protons and 8 neutrons. What is this atom's atomic mass number?

If an atom has a mass number of 20 and an atomic number of 8, how many neutrons are present in this atom?

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Suzanne Slade



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Atomic Structure Teacher's Guide Tina Youngman,2017

The Structure of Atoms Suzanne Slade,2006-08-15

Explains what an atom is and why it is important and describes the particles that make up atoms

Elementary Atomic

Structure Gordon Kemble Woodgate,1998

Chemistry for Kids, the Atomic Structure Lex Sharp,2018-03-24

This book is meant for students grades 6 and higher It addresses concepts that are critical to the understanding of the inner structure of the atom which lays the foundation for appreciating how elements behave in chemical reactions The text has been integrated with colorful images that help engage the visual with the conceptual No prerequisites are necessary The book is suitable for the beginner who has not studied chemistry before and for the student who wants to clarify questions for review purposes A couple of mathematical chapters were also infused into the book to support younger learners with the necessary mathematical background to succeed To get the most from this book students should summarize each chapter and rehearse the concepts Attempting to work independently through all problems at the end of the book is beneficial only before reading the solutions When solving problems one can be sure the chemical principles presented were well understood Restricting oneself to only reading is incomplete learning The series walks learners gradually and conceptually through the chemical and particle physics theories that are meant to jumpstart a more profound scientific understanding Topics covered in this volume Subatomic Particle Atomic Mass Period Table of Elements Atomic Number Isotope and Ion Notations Atomic Weight The Mole Molar Mass Exponents The Rule of Three Practice problems and easy to follow solutions

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