

Section Review 8.1

Part A Completion

- | | |
|----------------------|--------------|
| 1. compound | 6. low |
| 2. molecular | 7. high |
| 3. nonmetals | 8. atoms |
| 4. diatomic | 9. structure |
| 5. molecular formula | |

Part B True-False

- | | | |
|--------|--------|--------|
| 10. ST | 12. AT | 14. ST |
| 11. NT | 13. NT | |

Part C Matching

- | | | |
|-------|-------|-------|
| 15. e | 17. d | 19. c |
| 16. a | 18. b | |

Part D Questions and Problems

20. molecular compound
21. a. 4 carbon atoms, 10 hydrogen atoms
b. 6 carbon atoms, 5 hydrogen atoms, 1 fluorine atom
22. a. molecule d. atom
b. atom e. molecule
c. molecule

Section Review 8.2

Part A Completion

- stable electron
- covalent
- shared
- single
- unshared pairs
- double/triple
- coordinate covalent bond
- Energy
- bond dissociation energy
- resonance structure

Part B True-False

- | | | |
|--------|--------|--------|
| 11. NT | 13. AT | 15. ST |
| 12. NT | 14. AT | 16. NT |

Part C Matching

- | | | |
|-------|-------|-------|
| 17. e | 19. b | 21. d |
| 18. a | 20. c | |

Part D Questions and Problems

22. a. $\text{Br}:\text{Br}:$
- b. $\text{H}:\text{C}:::\text{N}:$
- c. $\begin{array}{c} \text{H} \\ | \\ \text{H}:\text{N}:\text{H} \\ | \\ \text{H} \end{array}$

Section 8.3

Part A Completion

- molecular orbitals
- bonding orbital
- lower
- sigma or σ
- pi or π
- three-dimensional
- VSEPR theory
- orbital hybridization

Part B True-False

- | | | |
|--------|--------|--------|
| 9. AT | 11. NT | 13. ST |
| 10. ST | 12. AT | 14. AT |

Part C Matching

- | | | |
|-------|-------|-------|
| 15. d | 17. a | 19. c |
| 16. e | 18. b | |

Part D Questions and Problems

20. Reading from left to right:
 sp^3 sp^2 sp^2 sp sp sp^3

Section 8.4

Part A Completion

- equally
- nonpolar
- unequally
- polar
- electronegativities
- dipole interactions
- hydrogen bond
- electronegative
- oxygen, nitrogen, or fluorine

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S Marginson



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