

Atomic Structure & Periodic Table

Complete the following table using the periodic table of elements.

| Isotope name | Atomic number | Mass number | Number of protons | Number of neutrons | Number of electrons |
|---------------|---------------|-------------|-------------------|--------------------|---------------------|
| Hydrogen - 2 | | | | | |
| | 2 | 4 | | | |
| | | | 4 | 5 | |
| | | 7 | 3 | | |
| Chlorine - 35 | | | | | |
| | 19 | | | 20 | |
| | | 24 | 12 | | |
| | | | 41 | 52 | |
| | | 107 | | | 47 |
| | 16 | | | 16 | |
| Uranium - 235 | | | | | |
| Scandium - 45 | | 45 | | 24 | |
| | | | 31 | 40 | 31 |
| | 46 | 106 | | | |
| Xenon - 136 | | | | | |

Atomic Structure The Periodic Table Answer Key

F Rizvi



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