

## Ideal gas law - problems and solutions

1. Ideal gases in a closed container initially have volume  $V$  and temperature  $T$ . The final temperature is  $5/4T$  and the final pressure is  $2P$ . What is the final volume of gases.

Known :

Initial volume ( $V_1$ ) =  $V$

Initial temperature ( $T_1$ ) =  $T$

Final temperature ( $T_2$ ) =  $5/4 T$

Initial pressure ( $P_1$ ) =  $P$

Final pressure ( $P_2$ ) =  $2P$

Wanted : Final volume ( $V_2$ )

Solution :

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

$$\frac{P V}{T} = \frac{(2P) V_2}{5/4 T}$$

$$\frac{V}{1} = \frac{2 V_2}{5/4}$$

$$5/4 V = 2 V_2$$

$$V_2 = \frac{5V}{4} \cdot \frac{2}{1}$$

$$V_2 = \frac{5V}{4} \times \frac{1}{2}$$

$$V_2 = \frac{5V}{8}$$

2. Determine the volume of 2.00 moles of gases (ideal gas) at STP. STP = Standard Temperature and Pressure.

Known :

Moles of gas ( $n$ ) = 2 moles

Standard temperature ( $T$ ) =  $0^\circ\text{C} = 0 + 273 = 273$  Kelvin

Standard pressure ( $P$ ) = 1 atm =  $1.013 \times 10^5$  Pa

Universal gas constant ( $R$ ) = 8.315 Joule/mole.Kelvin

Wanted : Volume of gases ( $V$ )

Solution :

Read more :

<https://physics.gurumuda.net/ideal-gas-law-problems-and-solutions.htm>

# Gas Law Problems With Solutions

**L Reisser**



## **Gas Law Problems With Solutions:**

## **Gas Law Problems With Solutions** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Gas Law Problems With Solutions**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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