

Question 2

Calculate the volumes of 100.00 ppm w/v solution needed to prepare the following solutions.

200 mL of 0.500 ppm w/v Fe: ✓ mL

100 mL of 1.00 ppm w/v Fe: ✓ mL

100 mL of 2.00 ppm w/v Fe: ✓ mL

100 mL of 3.00 ppm w/v Fe: ✓ mL

100 mL of 4.00 ppm w/v Fe: ✓ mL

Ppm Solution Preparation Formula

**Arthur L. Williams, Harland D.
Embree, Harold J. DeBey**

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